

# **Deciphering the isotopic imprint of nitrate to reveal nitrogen source and transport mechanisms in a tile-drained agroecosystem**

Yinchao Hu<sup>1</sup>, Zhongjie Yu<sup>1\*</sup>, Wendy H. Yang<sup>2,3</sup>, Andrew J. Margenot<sup>4</sup>, Lowell E. Gentry<sup>1</sup>, Michelle M. Wander<sup>1</sup>, Richard L. Mulvaney<sup>1</sup>, Corey A. Mitchell<sup>1</sup>, Carlos E. Guacho<sup>1</sup>

Departments of <sup>1</sup>Natural Resources and Environmental Sciences, <sup>2</sup>Plant Biology, <sup>3</sup>Earth Science and Environmental Change, and <sup>4</sup>Crop Sciences, University of Illinois at Urbana-Champaign, Urbana, Illinois, USA

\*Corresponding author: [zjyu@illinois.edu](mailto:zjyu@illinois.edu)

## **Abstract**

Installation of subsurface drainage systems has profoundly altered the nitrogen cycle in agricultural regions across the globe, facilitating substantial loss of nitrate (NO<sub>3</sub><sup>-</sup>) to surface water systems. Lack of understanding of the sources and processes controlling NO<sub>3</sub><sup>-</sup> loss from tile-drained agroecosystems hinders the development of management strategies aimed at reducing this loss. The natural abundance nitrogen and oxygen isotopes of NO<sub>3</sub><sup>-</sup> provide a valuable tool for differentiating nitrogen sources and tracking the biogeochemical transformations acting on NO<sub>3</sub><sup>-</sup>. This study combined multi-years of tile drainage measurements with NO<sub>3</sub><sup>-</sup> isotopic analysis to examine NO<sub>3</sub><sup>-</sup> source and transport mechanisms in a tile-drained corn-soybean field. The tile drainage NO<sub>3</sub><sup>-</sup> isotope data were supplemented by characterization of the nitrogen isotopic composition of potential NO<sub>3</sub><sup>-</sup> sources (fertilizer, soil nitrogen, and crop biomass) in the field and the oxygen isotopic composition of NO<sub>3</sub><sup>-</sup> produced by nitrification in soil incubations. The results show that NO<sub>3</sub><sup>-</sup> isotopes in tile drainage were highly responsive to tile discharge variation and fertilizer input. After accounting for isotopic fractionations during nitrification and denitrification, the isotopic signature of tile drainage NO<sub>3</sub><sup>-</sup> was temporally stable and similar to those of fertilizer and soybean residue during unfertilized periods. This temporal invariance in NO<sub>3</sub><sup>-</sup> isotopic signature indicates a nitrogen legacy effect, possibly resulting from N recycling at the soil microsite scale and a large water storage for NO<sub>3</sub><sup>-</sup> mixing. Collectively, these

results demonstrate how combining field  $\text{NO}_3^-$  isotope data with knowledge of isotopic fractionations can reveal mechanisms controlling  $\text{NO}_3^-$  cycling and transport under complex field conditions.

## **Plain Language Summary**

Installation of subsurface tile pipes in many poorly drained agricultural lands has facilitated a substantial loss of nitrate ( $\text{NO}_3^-$ ) to surface water systems. However, the nitrogen sources and related processes controlling  $\text{NO}_3^-$  export from tile-drained agricultural systems remain unclear. This study employed stable isotope techniques to investigate how  $\text{NO}_3^-$  is biologically produced and hydrologically transported in a tile-drained field. Stable isotopes are chemical variants of the same element and have long been used as a tracer of nitrogen cycling in environmental systems. By combining field measurements of  $\text{NO}_3^-$  isotopes in tile drainage with a detailed understanding of how these isotopes are altered by microbial reactions, we estimated the original isotope ratios of  $\text{NO}_3^-$  and compared these ratios to those of potential nitrogen sources in the field. The results show that the original isotope ratios of  $\text{NO}_3^-$  were similar to those of ammonia fertilizer and soybean biomass nitrogen and did not vary over time when there was no fertilizer input to the system. These findings indicate the presence of a large  $\text{NO}_3^-$  pool in the soil and a time lag between the moments when the source nitrogen was introduced into the system and when the  $\text{NO}_3^-$  was exported via tile drainage.

## **Highlights**

- The oxygen isotopic composition of nitrate produced by soil nitrification varied with the degree of soil nitrite accumulation.
- The dual isotopes of nitrate in tile drainage exhibited coupled variations and were highly responsive to variations in tile discharge.
- Combining field nitrate isotope data with the isotopic systematics of nitrification reveals a legacy effect controlling nitrate dynamics.

**Keywords:** tile drainage, nitrate isotopes, source partitioning, isotopic fractionation, nitrification, denitrification

## 1. Introduction

Chronic anthropogenic nitrogen (N) input into terrestrial and aquatic ecosystems exceeds established safe planetary boundaries, with detrimental consequences of biodiversity loss, water pollution, and deteriorating ecosystem resilience (Gruber and Galloway, 2008). In the U.S. Upper Midwest, approximately 40% of agricultural lands are currently drained by subsurface tiles to increase soil drainage for corn and soybean production (Castellano et al., 2019; Valayamkunnath et al., 2020). The installation of these drainage systems has profoundly altered regional hydrological and biogeochemical regimes, facilitating a substantial loss of nitrate ( $\text{NO}_3^-$ ) via tile discharge (Blann et al., 2009). Indeed, the combination of N fertilizer applications on intensively tile-drained watersheds has been considered the dominant source of riverine  $\text{NO}_3^-$  yields in the Mississippi River Basin (David et al., 2010).

Despite well-documented environmental impacts, the sources and processes sustaining high  $\text{NO}_3^-$  export from tile-drained agricultural landscapes remain unclear. For example, while labeled  $^{15}\text{N}$  tracer studies in the U.S. Corn Belt have consistently demonstrated that less than half of applied fertilizer N is recovered by the corn crop in the same year (Gardner and Drinkwater, 2009), it remains elusive how the majority of fertilizer N not recovered by the corn uptake is partitioned into various retention and loss pathways (Yan et al., 2020). Moreover, variations in  $\text{NO}_3^-$  load from many tile-drained fields and watersheds have been shown to scale linearly with precipitation and flow events (Bauwe et al., 2020; Danalatos et al., 2022), suggesting a biogeochemically stationary regime in  $\text{NO}_3^-$  export. This biogeochemical stationarity, often referred to as ‘chemostasis’ (Godsey et al., 2009), has been interpreted to indicate the presence of a large legacy N store resulting from  $\text{NO}_3^-$  accumulation in soil and groundwater systems and/or accretion and subsequent mineralization of soil organic N (Basu et al., 2010; Thompson et al., 2011; Van Meter et al., 2018). However, direct constraints and detailed mechanisms to corroborate this legacy N store in intensively managed agroecosystems are still lacking.

Enabled by methodological advances in the early 2000s, the last two decades have seen a bloom of studies that utilized the natural abundance  $^{15}\text{N}/^{14}\text{N}$  and  $^{18}\text{O}/^{16}\text{O}$  ratios of  $\text{NO}_3^-$  to investigate

reactive N in the environment. By convention,  $\text{NO}_3^-$  isotopes are reported using  $\delta$  notation, where  $\delta^{15}\text{N} = ([^{15}\text{N}/^{14}\text{N}]_{\text{sample}}/[^{15}\text{N}/^{14}\text{N}]_{\text{air}} - 1) \times 1000$  and  $\delta^{18}\text{O} = ([^{18}\text{O}/^{16}\text{O}]_{\text{sample}}/[^{18}\text{O}/^{16}\text{O}]_{\text{VSMOW}} - 1) \times 1000$ , in units of per mille (‰). As each isotope system offers complementary insights into  $\text{NO}_3^-$  sources and transformations (Fig. 1), dual analysis of  $\delta^{15}\text{N}_{\text{NO}_3}$  and  $\delta^{18}\text{O}_{\text{NO}_3}$  offers unique biogeochemical information unobtainable by bulk concentration measurement alone (Granger and Wankel, 2016). However, notwithstanding the increasing use of  $\text{NO}_3^-$  isotopes, there are persistent knowledge gaps in interpreting environmental  $\text{NO}_3^-$  isotope data for robust  $\text{NO}_3^-$  source characterization. For example, considerable uncertainties can result from simple mixing modeling of  $\text{NO}_3^-$  isotopes due to unknown or overlapping  $\text{NO}_3^-$  source endmembers (Kendall et al., 2007) (Fig. 1). Moreover, enzymatic bond breaking during microbial denitrification results in strong isotopic fractionation of  $\text{NO}_3^-$ , leading to substantial enrichment of heavier isotopes ( $^{15}\text{N}$  and  $^{18}\text{O}$ ) in the remaining  $\text{NO}_3^-$  (Mariotti et al., 1981). This denitrification isotope effect is expressed as a linear trajectory in the biplot of  $\delta^{15}\text{N}_{\text{NO}_3}$  and  $\delta^{18}\text{O}_{\text{NO}_3}$  (i.e., dual isotope space; Fig. 1), reflecting the coupling of N and O isotopic fractionations during the denitrification process (Granger et al., 2008; Boettger et al., 2022). While the expression of these isotopic fractionations can be a valuable tool for quantifying the occurrence and intensity of denitrification, the large isotopic enrichment resulting from denitrification obscures the original isotopic signature of  $\text{NO}_3^-$  and, therefore, requires careful corrections for N source partitioning using  $\text{NO}_3^-$  isotopes.

At the other end of the biogeochemical  $\text{NO}_3^-$  cycle, nitrification determines the initial isotopic imprint of  $\text{NO}_3^-$  produced from organic and ammoniacal N, which comprise the major anthropogenic and soil sources of  $\text{NO}_3^-$  in agricultural landscapes (Fig. 1). Previous studies of pure cultures and soil incubations have demonstrated a significant kinetic N isotope effect for the two-step nitrification of  $\text{NO}_3^-$  from ammonia ( $\text{NH}_3$ ) (Casciotti et al., 2003; Yu and Elliott, 2018). Consequently, when  $\text{NH}_3$  supply is high, such as after fertilizer applications, the  $\delta^{15}\text{N}_{\text{NO}_3}$  produced from nitrification is much lower than the  $\delta^{15}\text{N}$  of the source  $\text{NH}_3$ . For the  $\delta^{18}\text{O}_{\text{NO}_3}$  produced from nitrification (denoted as  $\delta^{18}\text{O}_{\text{NO}_3, \text{nit}}$ ), it is commonly assumed that, for the three O atoms of  $\text{NO}_3^-$  produced from bacterial nitrification, two O atoms are derived from water and one from  $\text{O}_2$  (Andersson and Hooper, 1983;

Kumar et al., 1983). Accordingly,  $\delta^{18}\text{O}_{\text{NO}_3,\text{nit}}$  has been construed to vary over the range of -10 to 10‰ (Fig. 1) (Kendall et al., 2007) given the  $\delta^{18}\text{O}$  of atmospheric  $\text{O}_2$  (23.5‰) (Kroopnick and Craig, 1972) and the normal  $\delta^{18}\text{O}$  range of environmental water (-25 to 4‰). However, recent studies of bacterial and archaeal pure cultures and freshwater assemblages have revealed a complex isotopic systematics of  $\delta^{18}\text{O}_{\text{NO}_3,\text{nit}}$  that includes kinetic fractionation during the incorporation of O atoms from water and  $\text{O}_2$  into the  $\text{NO}_3^-$  produced, as well as equilibrium fractionation occurring during the exchange of O atoms between water and nitrite ( $\text{NO}_2^-$ ), which is an intermediate of nitrification (Casciotti et al., 2010; Buchwald and Casciotti, 2010; Nishizawa et al., 2016; Boshers et al., 2019) (Inset in Fig. 1). Importantly, a robust predictive understanding of  $\delta^{18}\text{O}_{\text{NO}_3,\text{nit}}$  is vital to pinpointing the origin of the denitrification line in dual isotope space, where the corresponding value on the  $\delta^{15}\text{N}_{\text{NO}_3}$  axis represents the source signature of the nitrified  $\text{NO}_3^-$ , corrected for denitrification enrichment (Fig. 1). Despite its necessity for unbiased  $\text{NO}_3^-$  source partitioning, only one study (Snider et al., 2010) has thus far quantified O isotopic fractionations and exchange during soil nitrification. It remains unresolved whether isotopic systematics of  $\delta^{18}\text{O}_{\text{NO}_3,\text{nit}}$  based on pure cultures can be generalized to nitrification catalyzed by complex soil nitrifying communities and how changes in soil state variables (e.g., substrate availability) may influence the variability of  $\delta^{18}\text{O}_{\text{NO}_3,\text{nit}}$ .

This study presents results from a comprehensive investigation of  $\text{NO}_3^-$  isotopes in a tile-drained field under a corn-soybean rotation, representative of the U.S. Corn Belt. Given the potential for  $\text{NO}_3^-$  isotopes to summarize  $\text{NO}_3^-$  source and transformation mechanisms, we expected combining long-term tile drainage measurements with  $\text{NO}_3^-$  isotopic analysis to provide new insights into the sources and processes controlling tile drainage  $\text{NO}_3^-$  export. A unique aspect of this work is that the N isotopic composition of potential sources of  $\text{NO}_3^-$  in tile drainage, including fertilizer, soil N, and crop residues, were directly measured. Additionally, laboratory soil incubations were conducted to quantify isotope effects controlling  $\delta^{18}\text{O}_{\text{NO}_3,\text{nit}}$  under different nitrification conditions. These characterized source endmembers and isotope effects were then combined with  $\text{NO}_3^-$  isotopes measured in tile drainage to tease apart potentially intertwined mechanisms for  $\text{NO}_3^-$  transport and source variations. To our knowledge, this work represents the first attempt to employ such a fine-grained approach to

tackle  $\text{NO}_3^-$  sources and transport in terrestrial settings. We postulated that if tile drainage  $\text{NO}_3^-$  loss was directly controlled by management and biogeochemical processes, such as fertilizer input, crop rotation, and mineralization of soil organic N, the isotopic signature of  $\text{NO}_3^-$  after accounting for isotopic fractionations during nitrification and denitrification would reflect the dynamic source contributions. Additionally, the isotopic signature of  $\text{NO}_3^-$  would be temporally stable if  $\text{NO}_3^-$  in tile drainage was dominantly sourced from a large and well-mixed legacy N store in the soil-plant system.

## **2. Materials and Methods**

### **2.1. Site description and tile drainage management.**

The study field is located within the Upper Embarras River watershed near Tuscola, Illinois (39°43' N, 88°14' W). This site is representative of the recently glaciated Midwest, characterized by low-gradient topography and poorly drained soils classified as fine, mixed, superactive, mesic Typic Endoaquolls (Milford series) (USDA-NRCS, 2016). The system of parallel drains at this field was installed approximately 40 years ago at depths ranging from 1.2 to 1.5 m below the soil surface, and is composed of one main drain fed by a set of lateral drains spaced 30.5 m apart, all consisting of 12.7 cm diameter perforated plastic pipe (Fig. S1). Starting in 2015, each lateral tile was monitored for a replicated fertilizer management study using inline water level control structures (AgriDrain Corporation, USA) (Gentry et al., 2023).

The current study focuses on field-scale observations of three lateral tiles, hereafter referred to as Tiles A, B, and C (Fig. S1). Each study tile is 500 to 570 m in length and drains an area of approximately 1.7 ha. Identical agricultural management practices were applied to the soil overlying the three tiles. Briefly, a two-year crop rotation of corn-soybean, the dominant cropping scheme in the Upper Midwest, was used. During the period of this study (October 2020 to August 2023), corn was planted in 2021 and 2023, and soybean in 2022. One to 14 days prior to corn planting, anhydrous  $\text{NH}_3$  was knifed into the surface soil at a rate of 200 kg N  $\text{ha}^{-1}$ . No N fertilizer was applied to soybean. It is noteworthy that no  $\text{NO}_3^-$  fertilizer was applied to the study tiles between 2015 and 2023. This allows us to examine the extent to which a better accounting of nitrification and denitrification isotope effects can improve the isotopic fingerprinting of  $\text{NO}_3^-$ , without being affected by confounding effects

resulting from external  $\text{NO}_3^-$  input. Likewise, atmospheric  $\text{NO}_3^-$  deposition, which was approximately  $5 \text{ kg N ha}^{-1}$  during the study period (NADP, 2023), is deemed negligible compared to annual N input via fertilizer application and crop residue incorporation ( $> 400 \text{ kg N ha}^{-1}$ ). While soybean was no-till planted, the seedbed for corn was prepared by strip-tillage. As is typical in the region, the field is rainfed and not irrigated, so the only water input was precipitation.

Throughout the study period, tile discharge at each tile was measured using the inline control structure and a pressure transducer (see Yu et al. (2023) for more details). We collected weekly water samples from each of the three control structures when the tiles were flowing ( $n = 351$ ). These samples were vacuum filtered and frozen until concentration and isotopic ( $\text{NO}_3^-$  and water) analyses. Additionally, from April 2022 to August 2023, a custom-made bulk precipitation collector was deployed at the site to collect weekly cumulative precipitation samples for water isotope analysis.

## **2.2. Quantification of tile $\text{NO}_3^-$ load and $\text{NO}_3^-$ export regime.**

Daily  $\text{NO}_3^-$  loads in tile drainage were linearly extrapolated to estimate annual loads based on water year, defined as October 1 to September 31. The export regime of  $\text{NO}_3^-$  in tile drainage was quantified using the method of Godsey et al. (2009). For this analysis, measured  $\text{NO}_3^-$  concentrations and corresponding tile discharge were log-transformed, and the linear slope of the log-log regression of discharge and concentration was used to categorize  $\text{NO}_3^-$  export regime into chemostatic (slope  $\approx 0$ ) and non-chemostatic (slope significantly different than zero) patterns. To examine the effects of crop rotation and fertilizer input on  $\text{NO}_3^-$  export regime, we classified  $\text{NO}_3^-$  and tile discharge data by crop year (i.e., water year with corn or soybean being cultivated), and further divided the corn data into growing season and non-growing season based on the timing of spring N fertilizer application. Therefore, the corn non-growing season spans from the dormant period of a corn year until spring fertilizer application, followed by the corn growing season, which starts at fertilizer application and ends at corn harvest. This distinction of cropping phases is operationally defined and aligns well with the onset and cessation of tile drainage in the field (see below).

## **2.3. Determination of $\text{NO}_3^-$ source endmembers.**

We collected numerous surface soil samples (0-20 cm) from two locations at each study tile immediately following the application of anhydrous  $\text{NH}_3$  fertilizer in April 2021, targeting the area where  $\text{NH}_3$  was knife-injected. The collected samples were combined into six composite samples for determination of the  $\delta^{15}\text{N}$  of soil ammonium ( $\text{NH}_4^+$ ), which was used to represent the  $\delta^{15}\text{N}$  of applied  $\text{NH}_3$  fertilizer. Near the end of the growing seasons of 2021 and 2022 (i.e., at the R6 growth stage), we collected aboveground corn and soybean biomass for N isotopic analysis. Two 5 by 5 m microplots were established overlying each tile, offset from the center, for collection of eight corn plants in 2021 and 15 soybean plants in 2022 from each microplot. Biomass samples collected from each microplot were first separated into grain and stover and then respectively combined to generate composite samples. Additionally, we collected one soil core to a depth of 90 cm within each microplot following the crop biomass sampling in 2021 and 2022. These soil cores were sectioned at 0-15, 15-30, 30-60, and 60-90 cm depths for isotopic analyses of soil total N (TN).

#### **2.4. Laboratory nitrification experiment.**

Following soybean harvest in 2022, we sampled surface soil (0-30 cm) from two locations at each tile and combined all collected samples to generate a composite sample for a laboratory nitrification experiment. The physical and chemical properties of this composite soil are provided in Table S1. We slurred the soil for this experiment to provide a homogenized environment for characterizing integrated isotopic fractionations catalyzed by the soil microbial community (Taylor et al., 2019). The experimental protocol largely followed those developed for quantifying  $\delta^{18}\text{O}_{\text{NO}_3, \text{nit}}$  using nitrifier pure cultures (Casciotti et al., 2010; Buchwald and Casciotti, 2010). Specifically, we established two  $\text{NH}_4^+$  fertilization treatments, 100 mg N kg dry soil<sup>-1</sup> and 0 mg N kg dry soil<sup>-1</sup>, to quantify how  $\delta^{18}\text{O}_{\text{NO}_3, \text{nit}}$  values change with soil nitrification rate. The fertilized treatment was designed to mimic high soil  $\text{NH}_4^+$  conditions following field fertilizer applications (i.e., the initial concentration of 100 mg N kg dry soil<sup>-1</sup> is roughly equivalent to an application rate of 224 kg N ha<sup>-1</sup>).

For each fertilization treatment, we established three parallel  $\delta^{18}\text{O}_{\text{H}_2\text{O}}$  treatments, -6.8‰, 21.3‰, and 49.1‰, each with three replicates, to quantify O isotope exchange and fractionations during nitrification. These parallel treatments were established by mixing laboratory de-ionized water



and  $^{18}\text{O}$ -labeled water (10% atom percent; Cambridge Isotope Laboratory). To set up the incubation, replicate samples ( $n=72$  for the fertilized treatment and  $n=63$  for the non-fertilized treatment) were prepared by combining 8 grams of soil (dry weight equivalent) with 10 mL of de-ionized water in a 50-mL centrifuge tube and then preincubated for 24 h. To initiate the incubation, 30 mL of water containing different amounts of  $^{18}\text{O}$ -labeled water and a  $(\text{NH}_4)_2\text{SO}_4$  solution was added to each tube to achieve a 1:5 soil-to-water ratio and the desired  $\text{NH}_4^+$  fertilization and  $\delta^{18}\text{O}_{\text{H}_2\text{O}}$  levels. The soil slurries were then incubated at room temperature (21 °C) on a horizontal shaker and sampled at intervals of 0.5 to 14 h over a 72-h period. For each incubation interval, a set of replicate slurry were sampled by centrifuging at 2000 rpm for 10 min, and the supernatants were filtered through a sterile 0.2  $\mu\text{m}$  filter. We briefly opened the caps of remaining slurry samples every 24 h to ensure aerobic incubation conditions. We note that the  $^{18}\text{O}$ -labeled water contained a trace amount of background  $\text{NO}_3^-$  ( $\sim 0.1 \mu\text{g N L}^{-1}$ ). Preliminary measurements showed that this background  $\text{NO}_3^-$  to be at natural abundance with respect to  $^{18}\text{O}$  ( $\sim 20\text{‰}$ ) and thus had a negligible impact on the determination of  $\delta^{18}\text{O}_{\text{NO}_3,\text{nit}}$ ; however, the  $^{15}\text{N}$  enrichment exceeded 10% atom percent. Therefore,  $\delta^{15}\text{N}_{\text{NO}_3}$  results are only reported when there was no addition of  $^{18}\text{O}$ -labeled water (i.e., for the treatment with  $\delta^{18}\text{O}_{\text{H}_2\text{O}} = -6.8\text{‰}$ ).

## 2.5. Quantification of $\delta^{18}\text{O}_{\text{NO}_3,\text{nit}}$ and underlying isotopic fractionations and exchange

For both the fertilized and non-fertilized treatments of the nitrification experiment, we calculated  $\delta^{18}\text{O}_{\text{NO}_3,\text{nit}}$  using  $\text{NO}_3^-$  concentrations and  $\delta^{18}\text{O}_{\text{NO}_3}$  values measured at the beginning and end of the incubation. Therefore, these  $\delta^{18}\text{O}_{\text{NO}_3,\text{nit}}$  values represent the  $\delta^{18}\text{O}$  of net  $\text{NO}_3^-$  produced during the entire incubation period. To elucidate the fractionation mechanisms underlying  $\delta^{18}\text{O}_{\text{NO}_3,\text{nit}}$ , we adopted the established isotope model of bacterial and archaeal nitrification that considers kinetic and equilibrium O isotope effects for the two-step production of  $\text{NO}_3^-$  from  $\text{NH}_3$  via  $\text{NO}_2^-$  (Casciotti et al., 2010; Buchwald and Casciotti, 2010) (Fig. 1 insert):

$$\delta^{18}\text{O}_{\text{NO}_3,\text{nit}} = \frac{2}{3} \left\{ (1 - f_{\text{ex}}) \left[ \frac{1}{2} (\delta^{18}\text{O}_{\text{O}_2} + {}^{18}\epsilon_{\text{k},\text{O}_2}) + \frac{1}{2} (\delta^{18}\text{O}_{\text{H}_2\text{O}} + {}^{18}\epsilon_{\text{k},\text{H}_2\text{O},1}) \right] + f_{\text{ex}} (\delta^{18}\text{O}_{\text{H}_2\text{O}} + {}^{18}\epsilon_{\text{eq}}) \right\} + \frac{1}{3} (\delta^{18}\text{O}_{\text{H}_2\text{O}} + {}^{18}\epsilon_{\text{k},\text{H}_2\text{O},2}) \quad (\text{Eq. 1})$$

In Eq. 1,  $\delta^{18}\text{O}_{\text{O}_2}$  is the  $\delta^{18}\text{O}$  of  $\text{O}_2$ ,  $^{18}\epsilon_{\text{k},\text{O}_2}$  is the kinetic isotope effect for  $\text{O}_2$  incorporation,  $^{18}\epsilon_{\text{k},\text{H}_2\text{O},1}$  is the kinetic isotope effect for water incorporation during  $\text{NH}_3$  oxidation to  $\text{NO}_2^-$ ,  $^{18}\epsilon_{\text{k},\text{H}_2\text{O},2}$  is the kinetic isotope effect for water incorporation during  $\text{NO}_2^-$  oxidation to  $\text{NO}_3^-$ ,  $f_{\text{ex}}$  is the fraction of O atoms in  $\text{NO}_2^-$  that have exchanged with water, and  $^{18}\epsilon_{\text{eq}}$  is the equilibrium isotope fractionation factor for  $\text{NO}_2^-$  isotopic exchange. The O isotope effect associated with  $\text{NO}_2^-$  oxidation to  $\text{NO}_3^-$  ( $^{18}\epsilon_{\text{k},\text{NO}_2}$ ; Fig. 1 insert) is not included in Eq. 1 because this isotope effect was not expressed due to the near complete  $\text{NO}_2^-$  oxidation by the end of the incubation (see below). Eq. 1 can be rearranged to conform to a linear formulation of  $\delta^{18}\text{O}_{\text{NO}_3,\text{nit}}$  versus  $\delta^{18}\text{O}_{\text{H}_2\text{O}}$ :

$$\delta^{18}\text{O}_{\text{NO}_3,\text{nit}} = \left(\frac{2}{3} + \frac{1}{3}f_{\text{ex}}\right)\delta^{18}\text{O}_{\text{H}_2\text{O}} + \frac{1}{3}\left[(\delta^{18}\text{O}_{\text{O}_2} - ^{18}\epsilon_{\text{k},\text{O}_2} - ^{18}\epsilon_{\text{k},\text{H}_2\text{O},1})(1 - f_{\text{ex}}) - ^{18}\epsilon_{\text{k},\text{H}_2\text{O},2}\right] + \frac{2}{3}(f_{\text{ex}} ^{18}\epsilon_{\text{eq}}) \quad (\text{Eq. 2})$$

From Eq. 2,  $f_{\text{ex}}$  can be estimated using the slope of a linear regression of  $\delta^{18}\text{O}_{\text{NO}_3,\text{nit}}$  versus  $\delta^{18}\text{O}_{\text{H}_2\text{O}}$  across the three parallel  $\delta^{18}\text{O}_{\text{H}_2\text{O}}$  treatments. The intercept expression shown in Eq. 2 is a function of  $f_{\text{ex}}$ ,  $\delta^{18}\text{O}_{\text{O}_2}$ , and four O isotope effects ( $^{18}\epsilon_{\text{k},\text{O}_2}$ ,  $^{18}\epsilon_{\text{k},\text{H}_2\text{O},1}$ ,  $^{18}\epsilon_{\text{k},\text{H}_2\text{O},2}$ , and  $^{18}\epsilon_{\text{eq}}$ ). We combined  $^{18}\epsilon_{\text{k},\text{O}_2}$  and  $^{18}\epsilon_{\text{k},\text{H}_2\text{O},1}$  into a composite O isotope effect ( $^{18}\epsilon_{\text{comp}}$ ) to represent the lumped  $\delta^{18}\text{O}$  offset between  $\text{NO}_2^-$  produced from  $\text{NH}_3$  and its O substrates ( $\text{O}_2$  and water) (Casciotti et al., 2010). Furthermore, we assumed that  $^{18}\epsilon_{\text{comp}}$  and  $^{18}\epsilon_{\text{k},\text{H}_2\text{O},2}$  were constant between the fertilized and non-fertilized treatments. This allows us to solve for  $^{18}\epsilon_{\text{comp}}$  and  $^{18}\epsilon_{\text{k},\text{H}_2\text{O},2}$  using regression intercepts derived from the fertilized and non-fertilized incubations by further assuming a  $\delta^{18}\text{O}_{\text{O}_2}$  value of 23.5‰ and a  $^{18}\epsilon_{\text{eq}}$  value of 13‰ for both abiotic and enzyme-catalyzed O exchange (Casciotti et al., 2007; Buchwald and Casciotti, 2013) (i.e., solving two unknowns using a system of two equations).

## 2.6. $\text{NO}_3^-$ source characterization.

We estimated the source signature of tile drainage  $\text{NO}_3^-$  under in situ conditions (denoted as  $\delta^{15}\text{N}_{\text{source}}$ ) by combining measured tile drainage  $\text{NO}_3^-$  isotopes with the O isotopic systematics of  $\text{NO}_3^-$  quantified in the soil nitrification experiment. Specifically, we applied measured tile drainage  $\delta^{18}\text{O}_{\text{H}_2\text{O}}$  in the empirical relationships of  $\delta^{18}\text{O}_{\text{NO}_3,\text{nit}}$  versus  $\delta^{18}\text{O}_{\text{H}_2\text{O}}$  derived from the nitrification experiment to predict  $\delta^{18}\text{O}_{\text{NO}_3,\text{nit}}$  under in situ conditions. These predicted  $\delta^{18}\text{O}_{\text{NO}_3,\text{nit}}$  values were then compared to

measured  $\delta^{18}\text{O}_{\text{NO}_3}$  values in tile drainage to estimate the isotopic enrichment of  $\text{NO}_3^-$  resulting from denitrification. Subsequently, we derived  $\delta^{15}\text{N}_{\text{source}}$  by subtracting the estimated denitrification enrichment from measured  $\delta^{15}\text{N}_{\text{NO}_3}$  values. Graphically, this approach is essentially a linear projection in dual isotope space, where a measured data point is mapped to the predicted  $\delta^{18}\text{O}_{\text{NO}_3,\text{nit}}$  by following a denitrification trajectory with a slope of one.  $\delta^{15}\text{N}_{\text{source}}$  is then obtained as the corresponding value on the  $\delta^{15}\text{N}_{\text{NO}_3}$  axis (Fig. 1). Importantly, this method for estimating  $\delta^{15}\text{N}_{\text{source}}$  is dependent upon two critical assumptions: (1) the  $\delta^{18}\text{O}$  of water used by the soil nitrifiers is equal to that measured for tile drainage, and (2) denitrification fractionates  $\text{NO}_3^-$  isotopes along a linear trajectory with a slope of one in dual isotope space. These assumptions will be addressed before any conclusions are reached.

## **2.7. Chemical and isotopic analyses.**

Tile drainage and soil incubation samples were analyzed for  $\text{NO}_3^-$  and  $\text{NO}_2^-$  concentrations using a Dionex ICS1600 Ion Chromatograph. The  $\delta^{15}\text{N}$  and  $\delta^{18}\text{O}$  of  $\text{NO}_3^-$  were measured after conversion to  $\text{N}_2\text{O}$  by the denitrifier method (Weigand et al., 2016), using an Elementar Isoprime isotope ratio mass spectrometer (IRMS). Prior to  $\text{NO}_3^-$  isotopic analysis, any detectable  $\text{NO}_2^-$  was removed by decomposition using sulfamic acid (Granger and Sigman, 2009). International  $\text{NO}_3^-$  reference materials USGS34 and USGS35 and an internal  $\text{NO}_3^-$  working standard were used to calibrate  $\delta^{15}\text{N}_{\text{NO}_3}$  and  $\delta^{18}\text{O}_{\text{NO}_3}$  analyses, for which the precision was  $\pm 0.1\text{‰}$  in the former case and  $\pm 0.3\text{‰}$  in the latter. We measured the O and H isotopes of tile drainage water and on-site precipitation using wavelength-scanned cavity ring-down spectroscopy on a Picarro Model L2130-i. The spectroscopic measurements were made against laboratory reference materials calibrated to VSMOW, with a precision of  $\pm 0.1\text{‰}$  for  $\delta^{18}\text{O}_{\text{H}_2\text{O}}$  and  $\pm 0.5\text{‰}$  for  $\delta^2\text{H}$ . Ammonium and  $\text{NO}_3^-$  in the soil samples collected following anhydrous  $\text{NH}_3$  application were extracted using 2 M KCl and measured using a SmartChem 200 discrete flow analyzer. The  $\delta^{15}\text{N}$  value of extracted  $\text{NH}_4^+$  was determined by coupling the persulfate oxidation and the denitrifier method (Yu and Elliott, 2018). The collected soil core samples and crop biomass were oven-dried ( $60\text{ °C}$ ) for 6 days and finely ground for  $\delta^{15}\text{N}$  analysis using an elemental analyzer coupled with an IRMS.

## **2.8. Statistical analysis.**

Ordinary linear regression was used to examine the relationships between  $\delta^{15}\text{N}_{\text{NO}_3}$  and  $\delta^{18}\text{O}_{\text{NO}_3}$  in tile drainage and between  $\delta^{18}\text{O}_{\text{H}_2\text{O}}$  and  $\delta^{18}\text{O}_{\text{NO}_3,\text{nit}}$  measured in the nitrification experiment. We used analysis of covariance (ANCOVA) followed by Fisher's least significant difference procedure to detect significant differences in linear regression slope among groups. All statistical analyses were conducted using MATLAB and were evaluated at a significance level of 0.05.

### 3. Results

#### 3.1. Soil nitrification experiment.

Following  $\text{NH}_4^+$  fertilization,  $\text{NO}_3^-$  concentrations increased slowly over the first 36 hours of the incubation, followed by a much more rapid increase to peak concentrations of 94.2–99.1 mg N kg dry soil<sup>-1</sup> at 72 hours (Fig. 2a). The amounts of  $\text{NO}_3^-$  produced during the incubations were greater than 90% of the amended  $\text{NH}_4^+$  (i.e., 100 mg N kg dry soil<sup>-1</sup>) for all replicates, equivalent to a net  $\text{NO}_3^-$  production rate of approximately 32 mg N kg dry soil<sup>-1</sup> d<sup>-1</sup>.  $\text{NO}_2^-$  was detectable at the beginning of the incubation, reached maximum concentrations of 7.2–7.7 mg N kg dry soil<sup>-1</sup> between hours 48 and 58, and then decreased to less than 2 mg N kg dry soil<sup>-1</sup> at the end of the incubation. At that time,  $\text{NO}_2^-$  constituted less than 3% of the total  $\text{NO}_3^- + \text{NO}_2^-$  pool (Fig. 2a).  $\delta^{15}\text{N}_{\text{NO}_3}$  values decreased sharply to -22.1‰ following the onset of  $\text{NO}_3^-$  production, reaching a minimum of -24.2‰ at 14 hours (Fig. 2b). As  $\text{NO}_3^-$  production proceeded,  $\delta^{15}\text{N}_{\text{NO}_3}$  values gradually increased and eventually approached the initial  $\delta^{15}\text{N}$  of the amended  $\text{NH}_4^+$  (i.e., 0.1‰). This temporal progression of  $\delta^{15}\text{N}_{\text{NO}_3}$  can be well explained by a kinetic N isotope effect for  $\text{NH}_4^+$  oxidation, which was estimated to be 25.3‰ by fitting a closed-system Rayleigh product equation (Fig. S3).

Compared to the fertilized treatment, the net rates of  $\text{NO}_3^-$  production were much lower under the non-fertilized treatment (Fig. 2d), and the amount of  $\text{NO}_3^-$  produced over the entire incubation ranged between 4.3 and 5.8 mg N kg dry soil<sup>-1</sup>. During the net  $\text{NO}_3^-$  production,  $\delta^{15}\text{N}_{\text{NO}_3}$  values increased from 4.8‰ to 7.2‰ by hour 12 and then gradually decreased to 3.2‰ at the end of the incubation (Fig. 2e). This steady decline in  $\delta^{15}\text{N}_{\text{NO}_3}$  observed after hour 12 corresponds to a  $\delta^{15}\text{N}$  value of net produced  $\text{NO}_3^-$  of -0.8‰.  $\text{NO}_2^-$  was not detectable throughout the non-fertilized incubation.

The temporal progression of  $\delta^{18}\text{O}_{\text{NO}_3}$  values was similar in the fertilized and non-fertilized treatments (Fig. 2c and 2f). Specifically,  $\delta^{18}\text{O}_{\text{NO}_3}$  values varied among the three  $\delta^{18}\text{O}_{\text{H}_2\text{O}}$  treatments at the onset of  $\text{NO}_3^-$  production, reflecting mixing of newly produced  $\text{NO}_3^-$  with the background pool of soil  $\text{NO}_3^-$ . With fertilization,  $\delta^{18}\text{O}_{\text{NO}_3}$  values evolved gradually, obtaining final values of -2.2 to 45.2‰ for the three  $\delta^{18}\text{O}_{\text{H}_2\text{O}}$  treatments. In comparison to the fertilized treatment, the final  $\delta^{18}\text{O}_{\text{NO}_3}$  values without fertilization varied over a narrower range of -3.4 to 28.8‰. The estimated  $\delta^{18}\text{O}_{\text{NO}_3,\text{nit}}$  values correlated linearly with  $\delta^{18}\text{O}_{\text{H}_2\text{O}}$  across the three parallel  $\delta^{18}\text{O}_{\text{H}_2\text{O}}$  treatments (Fig. 3), giving a slope ( $\pm 1$  SE) of  $0.86 \pm 0.00$  and an intercept ( $\pm 1$  SE) of  $3.56 \pm 0.05$  for the fertilized treatment, while the corresponding values for the non-fertilized treatment were  $0.67 \pm 0.01$  and  $-1.62 \pm 0.21$ . Based on Eq. 2 and the estimated regression slopes,  $f_{\text{ex}}$  was estimated to be  $59 \pm 0\%$  and  $0 \pm 3\%$  for the fertilized and non-fertilized treatments, respectively. For both fertilized and non-fertilized treatments,  $^{18}\epsilon_{\text{comp}}$  and  $^{18}\epsilon_{\text{k,H}_2\text{O},2}$  were estimated using the regression intercepts to be  $24.3 \pm 1.0\%$  and  $4.5 \pm 0.4\%$ , respectively.

### 3.2. Tile discharge and $\text{NO}_3^-$ concentrations.

The quantity and duration of tile discharge in the field varied interannually during the study period (Fig. 4). Each water year, tile flow ceased in the early peak growing season (i.e., late July to early August) due to enhanced crop evapotranspiration. The onset of tile discharge was more variable and largely dependent on triggering precipitation events following crop harvest. On average across all three tiles, the total tile discharge ( $\pm 1$  SD) was  $303 \pm 14$  mm,  $324 \pm 13$  mm, and  $217 \pm 14$  mm for the three years, respectively. The annual  $\text{NO}_3^-$  loads ( $\pm 1$  SD) were higher in the 2021 corn year and the 2022 soybean year ( $25.8 \pm 3.7$  and  $26.3 \pm 3.4$  kg N ha<sup>-1</sup>, respectively) than in the 2023 corn year ( $21.1 \pm 2.6$  kg N ha<sup>-1</sup>).

Tile  $\text{NO}_3^-$  concentrations were more variable during the two corn years (2021 and 2023), as compared to the 2022 soybean year (Fig. 4b). Specifically, while  $\text{NO}_3^-$  concentrations increased at a slow rate during the corn non-growing season in 2021 and 2023, dramatic increases in  $\text{NO}_3^-$  concentrations were observed in tile drainage following fertilizer input. This was particularly the case following fertilizer application in 2021, where consecutive precipitation events triggered extended tile flows and a substantial loss of  $\text{NO}_3^-$  in tile drainage. Among the three tiles, Tile A had consistently

lower  $\text{NO}_3^-$  concentrations, possibly due to fine-scale soil property variation as this tile intersects more depression areas than the other two tiles (Fig. S1). For all three cropping phases (i.e., corn growing season, corn non-growing season, and soybean year) and three individual tiles, the log-log slope of  $\text{NO}_3^-$  concentration versus tile discharge was essentially zero (i.e., absolute magnitude less than 0.04), and none of the log-log regression fits was statistically significant (Fig. 5 left column). These results indicate a persistent chemostasis in tile  $\text{NO}_3^-$  export.

### 3.3. Variations in $\text{NO}_3^-$ isotopes in tile drainage.

$\text{NO}_3^-$  isotopes in tile drainage varied distinctly with the cropping phases (Fig. 4c and 4d). Both  $\delta^{15}\text{N}_{\text{NO}_3}$  and  $\delta^{18}\text{O}_{\text{NO}_3}$  values declined steadily by 4-5‰ during the corn non-growing season in 2021 and 2023. Following fertilizer application and at the initiation of the corn growing season in 2021,  $\delta^{15}\text{N}_{\text{NO}_3}$  values declined sharply to as low as -5‰. However, this period of low  $\delta^{15}\text{N}_{\text{NO}_3}$  was short-lived, as  $\delta^{15}\text{N}_{\text{NO}_3}$  values increased rapidly with declining tile discharge toward the peak growing season. As documented by Fig. 4c, a similar response of  $\delta^{15}\text{N}_{\text{NO}_3}$  was also observed following fertilizer application in 2023, albeit to a lesser extent due to drier conditions in the early summer of 2023 (Fig. 4a). Compared to  $\delta^{15}\text{N}_{\text{NO}_3}$  values,  $\delta^{18}\text{O}_{\text{NO}_3}$  values exhibited a steeper increase during the corn growing season, and a strong response to fertilizer input was generally lacking (Fig. 4d). Tile drainage  $\delta^{15}\text{N}_{\text{NO}_3}$  and  $\delta^{18}\text{O}_{\text{NO}_3}$  values measured during the soybean year varied over relatively narrower ranges and were generally similar to those measured at the end of the 2021 corn year.

In contrast to tile drainage  $\text{NO}_3^-$  concentration, both  $\delta^{15}\text{N}_{\text{NO}_3}$  and  $\delta^{18}\text{O}_{\text{NO}_3}$  values were highly responsive to tile discharge variation. This was especially the case during the corn non-growing season, when lower  $\delta^{15}\text{N}_{\text{NO}_3}$  and  $\delta^{18}\text{O}_{\text{NO}_3}$  values were consistently observed during high flow events (i.e., tile discharge  $> 1 \text{ mm d}^{-1}$ ; Fig. 5). This high sensitivity to flow event was confirmed by a significant and negative linear relationship between the two isotopes and the logarithm of tile discharge across three cropping phases and monitored tiles (Fig. 5 middle and right columns).

Plotting the measured  $\delta^{15}\text{N}_{\text{NO}_3}$  and  $\delta^{18}\text{O}_{\text{NO}_3}$  values revealed significant yet variable linear relationships in dual isotope space (Fig. 6). Specifically,  $\delta^{15}\text{N}_{\text{NO}_3}$  and  $\delta^{18}\text{O}_{\text{NO}_3}$  values measured during the corn non-growing season and soybean year appeared to follow a similar linear trajectory, while

those measured during the corn growing season were more scattered due to low  $\delta^{15}\text{N}_{\text{NO}_3}$  values observed following fertilizer applications. Results from an ANCOVA confirmed that the slope of the  $\delta^{15}\text{N}_{\text{NO}_3}$ -versus- $\delta^{18}\text{O}_{\text{NO}_3}$  relationship was not significantly different between the corn non-growing season and soybean year for Tiles B and C but was higher in either of these two phases without N fertilizer than in the corn growing season (Table S2). For Tile A, the slope was not significantly different among the three cropping phases (Table S2). Therefore, for consistency across the three tiles, we fit two regression lines to  $\text{NO}_3^-$  isotopes measured from each tile to best characterize the linear coupling of  $\delta^{15}\text{N}_{\text{NO}_3}$  and  $\delta^{18}\text{O}_{\text{NO}_3}$  values – one regression for all measurements and one regression for measurements during the corn non-growing season and soybean year only. The slopes of the regression lines based on the corn non-growing season and soybean year ranged from 0.90 to 1.06 across the three tiles (Fig. 6; Table S2). Including isotope data measured during the corn growing season significantly lowered the estimated slopes from near one to a range of 0.58 to 0.64 for the three tiles (Fig. 6; Table S2).

### 3.4. Water isotopes in tile drainage.

The  $\delta^{18}\text{O}_{\text{H}_2\text{O}}$  values of tile discharge varied between -7.4‰ and -5.5‰ throughout the study period (Fig. 4e). Variations in the  $\delta^{18}\text{O}_{\text{H}_2\text{O}}$  of tile discharge were highly dampened with respect to the  $\delta^{18}\text{O}_{\text{H}_2\text{O}}$  of precipitation, which varied between -16.6‰ and 0.7‰. Nevertheless, some synchronicities between the  $\delta^{18}\text{O}_{\text{H}_2\text{O}}$  values of precipitation and tile discharge were evident in the period for which contemporaneous tile discharge and precipitation measurements were available. Moreover, during individual flow events, the  $\delta^{18}\text{O}_{\text{H}_2\text{O}}$  of tile discharge can often be characterized by transient fluctuations that were correlated with the  $\delta^{18}\text{O}_{\text{H}_2\text{O}}$  of precipitation (Fig. 4e). Plotting the  $\delta^{18}\text{O}_{\text{H}_2\text{O}}$  and  $\delta^2\text{H}$  values of tile drainage revealed a close alignment with the local meteoric water line based on the precipitation isotopes (Fig. S2), indicating that water isotopes in tile drainage were unaffected by evaporative enrichment.

### 3.5. $\text{NO}_3^-$ source endmembers.

The  $\delta^{15}\text{N}$  value ( $\pm 1$  SD) of soil  $\text{NH}_4^+$  collected following anhydrous  $\text{NH}_3$  application was  $-0.6 \pm 1.9\text{‰}$  (Fig. 7). The  $\delta^{15}\text{N}$  values of soil TN did not differ between the two sampling years, and the average

values for the four soil layers varied over a narrow range of 5.1 to 5.5‰. Based on all soil core samples collected, the concentration-weighted average  $\delta^{15}\text{N}$  value of soil TN was  $5.3 \pm 0.7\text{‰}$ . Using  $\delta^{15}\text{N}$  values determined for corn grain ( $5.3 \pm 0.3\text{‰}$ ) and stover ( $7.4 \pm 0.7\text{‰}$ ), a weighted sum of  $5.8 \pm 0.3\text{‰}$  was obtained for total corn aboveground biomass N. The  $\delta^{15}\text{N}$  values of soybean grain ( $-0.2 \pm 0.1\text{‰}$ ) and stover ( $-0.3 \pm 0.4\text{‰}$ ) were indistinguishable, and the  $\delta^{15}\text{N}$  value of total soybean biomass N was estimated to be  $-0.2 \pm 0.1\text{‰}$  (Fig. 7).

### 3.6. $\text{NO}_3^-$ source characterization.

Given the distinct responses of  $\text{NO}_3^-$  isotopes to different cropping phases (Fig. 4), we calculated  $\text{NO}_3^-$  load-weighted mean values of  $\delta^{15}\text{N}_{\text{NO}_3}$  and  $\delta^{18}\text{O}_{\text{NO}_3}$  for each cropping phase to estimate  $\delta^{15}\text{N}_{\text{source}}$ . Across all three cropping phases and three tiles, these values ranged from 1.9‰ to 11.2‰ for  $\delta^{15}\text{N}_{\text{NO}_3}$  and from 3.2‰ to 8.0‰ for  $\delta^{18}\text{O}_{\text{NO}_3}$  (Fig. 7). To account for the effect of variable nitrification rate on  $\delta^{18}\text{O}_{\text{NO}_3, \text{nit}}$  and thusly on  $\delta^{15}\text{N}_{\text{source}}$ , we predicted  $\delta^{18}\text{O}_{\text{NO}_3, \text{nit}}$  as a range of values using the two  $\delta^{18}\text{O}_{\text{NO}_3, \text{nit}}$ -versus- $\delta^{18}\text{O}_{\text{H}_2\text{O}}$  relationships derived from the fertilized and non-fertilized soil incubations (Fig. 3). For illustrative purposes, in Fig. 7, we show source characterization results using the  $\text{NO}_3^-$  load-weighted mean  $\delta^{18}\text{O}_{\text{H}_2\text{O}}$  value of tile drainage for all three tiles (i.e.,  $-6.3\text{‰}$ ). In this case,  $\delta^{18}\text{O}_{\text{NO}_3, \text{nit}}$  was predicted to range between  $-5.8\text{‰}$  and  $-1.9\text{‰}$  using the two derived  $\delta^{18}\text{O}_{\text{NO}_3, \text{nit}}$ -versus- $\delta^{18}\text{O}_{\text{H}_2\text{O}}$  relationships (horizontal lines in Fig. 7). Based on this range of  $\delta^{18}\text{O}_{\text{NO}_3, \text{nit}}$ ,  $\delta^{15}\text{N}_{\text{source}}$  values were estimated to vary between  $-8.4\text{‰}$  and  $1.3\text{‰}$  when combining all three tiles and cropping phases (grey shaded area in Fig. 7).

Source characterization results based on  $\delta^{18}\text{O}_{\text{H}_2\text{O}}$  values specific to each tile and cropping phase are shown in Table 1. When considering all three tiles collectively,  $\delta^{15}\text{N}_{\text{source}}$  values were estimated to vary over a range of  $-8.3\text{‰}$  to  $-3.0\text{‰}$  for the corn growing season,  $-4.4\text{‰}$  to  $0.8\text{‰}$  for the corn non-growing season, and  $-4.8$  to  $1.4\text{‰}$  for the soybean year (Table 1). Notably,  $\delta^{15}\text{N}_{\text{source}}$  values were not significantly different between the corn non-growing season and soybean year as indicated by the overlapping ranges, whereas  $\delta^{15}\text{N}_{\text{source}}$  values estimated for the corn growing season were generally lower.

## 4. Discussion



#### 4.1. Oxygen isotope systematics of $\text{NO}_3^-$ produced by soil nitrification

The results from the nitrification experiment, which revealed a linear dependence of  $\delta^{18}\text{O}_{\text{NO}_3, \text{nit}}$  on  $\delta^{18}\text{O}_{\text{H}_2\text{O}}$ , are consistent with previous pure culture studies (Casciotti et al., 2010; Buchwald and Casciotti, 2010). It appears that this dependence on  $\delta^{18}\text{O}_{\text{H}_2\text{O}}$  varied with  $\text{NH}_4^+$  availability and nitrification rate, as indicated by the significantly different slope and intercept of the  $\delta^{18}\text{O}_{\text{NO}_3, \text{nit}}$ -versus- $\delta^{18}\text{O}_{\text{H}_2\text{O}}$  relationship between the fertilized and non-fertilized treatments (Fig. 3). The range of  $f_{\text{ex}}$  (0-59%) derived based on a single soil in this study falls within the range reported by Snider et al. (2010) for three different agricultural and forest soils (37-88%) but is wider than those observed in  $\text{NH}_3$  and  $\text{NO}_2^-$  oxidizing cultures grown under conditions relevant to marine nitrification (0-30%) (Casciotti et al., 2010; Buchwald et al., 2012).

The observed variability of  $\delta^{18}\text{O}_{\text{NO}_3, \text{nit}}$  and  $f_{\text{ex}}$  and their dependence on  $\text{NH}_4^+$  availability and nitrification rates can be explained by the accumulation of  $\text{NO}_2^-$  as a nitrification intermediate. Specifically, high  $\text{NH}_4^+$  availability in the fertilized treatment would be expected to stimulate the growth of  $\text{NH}_3$  oxidizing bacteria/archaea, resulting in increased intra- and extracellular  $\text{NO}_2^-$  concentrations and acidic pH in the cellular periplasm (Andersson et al., 1982). This combination of high  $\text{NO}_2^-$  abundance and low pH may accelerate both abiotic exchange between  $\text{NO}_2^-$  and water and biologically mediated exchange occurring during  $\text{NH}_3$  oxidation (Casciotti et al., 2007; Boshers et al., 2019). Consequently, the substantial exchange between  $\text{NO}_2^-$  and water under the fertilized condition resulted in a higher sensitivity of  $\delta^{18}\text{O}_{\text{NO}_3, \text{nit}}$  to variations in  $\delta^{18}\text{O}_{\text{H}_2\text{O}}$ , as reflected by the higher slope for the  $\delta^{18}\text{O}_{\text{NO}_3, \text{nit}}$ -versus- $\delta^{18}\text{O}_{\text{H}_2\text{O}}$  relationship (Fig. 3). Moreover, the equilibrium isotope effect for  $\text{NO}_2^-$  isotopic exchange (as represented by  $^{18}\epsilon_{\text{eq}}$ ) dictates that at equilibrium, the  $\delta^{18}\text{O}$  of  $\text{NO}_2^-$  is higher than  $\delta^{18}\text{O}_{\text{H}_2\text{O}}$  by 13‰. The propagation of the impact of this equilibrium fractionation into the  $\text{NO}_3^-$  produced can explain why final  $\delta^{18}\text{O}_{\text{NO}_3}$  values were higher for the fertilized than non-fertilized incubation, which likewise led to a larger intercept of the  $\delta^{18}\text{O}_{\text{NO}_3, \text{nit}}$ -versus- $\delta^{18}\text{O}_{\text{H}_2\text{O}}$  relationship (Fig. 3). On the other hand, nitrification in the absence of ammoniacal fertilization was limited by  $\text{NH}_3$  production from soil organic N (ammonification). This substrate limitation facilitated a tighter coupling of  $\text{NH}_3$  and  $\text{NO}_2^-$  oxidation and consequently prevented  $\text{NO}_2^-$  accumulation and  $\text{NO}_2^-$

isotopic exchange. Importantly,  $\text{NO}_2^-$  accumulation is frequently reported for agricultural soils after N fertilizer applications (Jones and Hedlin, 1970; Smith et al., 1997; Venterea, 2007; Maharjan and Venterea, 2013). This accumulation can be attributed to a delay in protein synthesis by  $\text{NO}_2^-$  oxidizing bacteria in response to increased  $\text{NH}_3$  oxidation rates (Giguere et al., 2018) or to a kinetic constraint that requires moderate  $\text{NO}_2^-$  accumulation to balance the rates of  $\text{NH}_3$  and  $\text{NO}_2^-$  oxidation (Taylor et al., 2019). Therefore, it can be assumed that the variable  $f_{\text{ex}}$  and  $\delta^{18}\text{O}_{\text{NO}_3,\text{nit}}$  observed in our soil incubations are representative of those mediated by nitrifier communities in agricultural soil, where large variation in substrate availability is common following fertilizer application.

Our estimate of  $^{18}\epsilon_{\text{comp}}$  ( $24.3 \pm 1.0\text{‰}$ ) is well within the range previously reported for  $\text{NH}_3$ -oxidizing bacteria and archaea (18-38‰) (Casciotti et al., 2010; Nishizawa et al., 2016), whereas the estimated  $^{18}\epsilon_{\text{k,H}_2\text{O},2}$  ( $4.5 \pm 0.4\text{‰}$ ) is lower than the reported range for  $\text{NO}_2^-$ -oxidizing bacteria (9-25‰) (Buchwald and Casciotti, 2010). We are unable to resolve whether the lower  $^{18}\epsilon_{\text{k,H}_2\text{O},2}$  from our soil incubations is due to an intrinsic difference in fractionation behavior between bacterial pure cultures and the presumably more complex nitrifier community present in soils at the study site, or is due to our assumption of constant  $^{18}\epsilon_{\text{comp}}$  and  $^{18}\epsilon_{\text{k,H}_2\text{O},2}$  between fertilized and non-fertilized treatments required to solve for these two terms. Regardless, the positive identification of  $^{18}\epsilon_{\text{comp}}$  and  $^{18}\epsilon_{\text{k,H}_2\text{O},2}$  is in line with the consensus reached in pure culture studies that enzyme-mediated extractions of O atoms from water and  $\text{O}_2$  and subsequent incorporations into nitrification-produced  $\text{NO}_3^-$  are associated with large kinetic fractionations.

Based on the current understanding of the fractionation mechanisms underlying  $\delta^{18}\text{O}_{\text{NO}_3,\text{nit}}$ , Boshers et al. (2019) conducted a numerical simulation to show that  $\delta^{18}\text{O}_{\text{NO}_3,\text{nit}}$  can be largely constrained to be within  $\pm 3\text{‰}$  of the corresponding  $\delta^{18}\text{O}_{\text{H}_2\text{O}}$ . In our case, when using the  $\text{NO}_3^-$  load-weighted mean  $\delta^{18}\text{O}_{\text{H}_2\text{O}}$  of tile drainage ( $-6.3\text{‰}$ ),  $\delta^{18}\text{O}_{\text{NO}_3,\text{nit}}$  was predicted to range between  $-5.8\text{‰}$  and  $-1.9\text{‰}$ . Therefore, our results support the conclusion of Boshers et al. (2019) and advocate a homologous model of  $\delta^{18}\text{O}_{\text{NO}_3,\text{nit}}$  for both nitrifying pure cultures and soil systems. These results also challenge the simplifying assumption adopted by many recent watershed studies that  $\delta^{18}\text{O}_{\text{NO}_3,\text{nit}}$  is solely dependent on the  $\delta^{18}\text{O}$  values of  $\text{O}_2$  and water and the proportional contribution of these two O

sources during nitrification. More studies are needed to quantify  $\delta^{18}\text{O}_{\text{NO}_3, \text{nit}}$  in soils from a broad range of environments and to more completely understand the variability of  $\delta^{18}\text{O}_{\text{NO}_3, \text{nit}}$  under varying thermodynamic and kinetic conditions.

#### **4.2. Source and transport mechanisms for tile drainage $\text{NO}_3^-$ export.**

The tile drainage measurements revealed marked chemostasis in  $\text{NO}_3^-$  export across all three cropping phases at our study site (Fig. 5 left column). The observation that variation in tile  $\text{NO}_3^-$  loads was predominantly determined by changes in tile discharge rather than  $\text{NO}_3^-$  concentration indicates transport-limited systems and the likely presence of a large  $\text{NO}_3^-$  store that can be readily mobilized in proportion to flow generation (Basu et al., 2010; Thompson et al., 2011). This interpretation is conceptually supported by the isotopic composition of  $\text{NO}_3^-$  in tile drainage measured during the corn non-growing season and soybean year. Specifically, variations in  $\delta^{15}\text{N}_{\text{NO}_3}$  and  $\delta^{18}\text{O}_{\text{NO}_3}$  observed during these periods were strongly coupled (Fig. 6). Moreover, among all potential  $\text{NO}_3^-$  sinks measured (i.e., soil TN and crop biomass), only soybean biomass had  $\delta^{15}\text{N}$  values lower than the temporally integrated  $\delta^{15}\text{N}$  values of tile drainage  $\text{NO}_3^-$  (Fig. 7), which can be well explained by biological  $\text{N}_2$  fixation carried out by the soybean plants. These observations point to denitrification, rather than  $\text{NO}_3^-$  assimilation by soil microbes or crop  $\text{NO}_3^-$  uptake, as the dominant process regulating the coupling and enrichment of  $\delta^{15}\text{N}_{\text{NO}_3}$  and  $\delta^{18}\text{O}_{\text{NO}_3}$  in tile drainage. Consequently, the observed strong coupling of  $\delta^{15}\text{N}_{\text{NO}_3}$  and  $\delta^{18}\text{O}_{\text{NO}_3}$  during unfertilized periods implies that tile drainage  $\text{NO}_3^-$  had an approximately constant  $\delta^{15}\text{N}_{\text{source}}$  value after accounting for denitrification enrichment and was not the product of dynamic mixing of multiple  $\text{NO}_3^-$  sources with distinct isotopic composition. These results support our assertion that the presence of a large and well-mixed  $\text{NO}_3^-$  store will result in a temporally invariant source signature of  $\text{NO}_3^-$  in tile drainage.

The estimated tile drainage  $\delta^{15}\text{N}_{\text{source}}$  values for the corn growing season and soybean year ranged between -4.4‰ and 1.4‰ across the three study tiles (Table 1). These values were similar to the  $\delta^{15}\text{N}$  values of applied fertilizer  $\text{NH}_3$  and soybean biomass but markedly different than soil TN and corn biomass N values (Fig. 7). This comparison implies the potential role of fertilizer- and/or soybean residue-derived N as a persistent source of tile drainage  $\text{NO}_3^-$  loss, even during the soybean

year, when no N fertilizer or soybean residue was input to the soil. In contrast, the  $\delta^{15}\text{N}_{\text{source}}$  values estimated for the corn growing season (-8.3 to -3.0‰) were lower than the  $\delta^{15}\text{N}$  of all potential N sources in our system (Fig. 7), which is consistent with the characteristically low  $\delta^{15}\text{N}_{\text{NO}_3}$  values observed following fertilizer applications (Fig. 4c) and can only be explained by the kinetic N isotope effect resulting from partial nitrification of applied fertilizer  $\text{NH}_3$  (Fig. 2b).

To test the sensitivity of the estimated  $\delta^{15}\text{N}_{\text{source}}$  values to water sources used during nitrification, we explored the possibility that soil nitrification mainly occurred in the aerobic surface soil so that the  $\delta^{18}\text{O}$  of water used in in-situ nitrification was subject to seasonal variations with the  $\delta^{18}\text{O}_{\text{H}_2\text{O}}$  of precipitation. We show in Table S3 that  $\delta^{15}\text{N}_{\text{source}}$  values remain essentially unchanged when using temporally integrated precipitation  $\delta^{18}\text{O}_{\text{H}_2\text{O}}$  values based on the three cropping phases (i.e., -7.3 to -5.1‰) to predict  $\delta^{18}\text{O}_{\text{NO}_3, \text{nit}}$ . This confirms that the use of tile drainage  $\delta^{18}\text{O}_{\text{H}_2\text{O}}$  values is robust. Additionally, the estimated  $\delta^{15}\text{N}_{\text{source}}$  values are conditioned on the assumption that denitrification exerts an identical fractionation on the N and O isotopes of  $\text{NO}_3^-$ , such that the slope of denitrification trajectory in dual isotope space is equal to one. This assumption is supported by the approximately 1:1 coupling of  $\delta^{15}\text{N}_{\text{NO}_3}$  and  $\delta^{18}\text{O}_{\text{NO}_3}$  observed during the corn non-growing season and soybean year (Fig. 6). It is also consistent with the intrinsic N and O isotopic fractionations catalyzed by the *Nar* dissimilatory  $\text{NO}_3^-$  reductase (Granger et al., 2008; Boettger et al., 2022), which is considered the dominant respiratory pathway for heterotrophic  $\text{NO}_3^-$  consumption in freshwater systems (Granger and Wankel, 2016). Interestingly, if we include  $\text{NO}_3^-$  isotope data measured during the corn growing season in the linear regression of  $\delta^{15}\text{N}_{\text{NO}_3}$ -versus- $\delta^{18}\text{O}_{\text{NO}_3}$ , the regression slope is lowered to be around 0.6 (Fig. 6; Table S2). As the  $\delta^{15}\text{N}_{\text{NO}_3}$  values measured following fertilizer applications were affected by the kinetic isotope effect for nitrification, this change in the regression slope provides direct evidence for the isotopic overprinting of nitrification on denitrification (Granger and Wankel, 2016). Extending this finding to surface- and groundwater systems, the commonly observed denitrification slopes being lower than one in these systems may partially result from temporal and spatial aggregations of  $\text{NO}_3^-$  variably fractionated during nitrification and denitrification.

Therefore, the estimated  $\delta^{15}\text{N}_{\text{source}}$  values across all cropping phases depict a bimodal pattern in  $\text{NO}_3^-$  source production, with fertilizer applications triggering a pulsed release of  $\text{NO}_3^-$  that shifts  $\delta^{15}\text{N}_{\text{source}}$  in tile drainage from an isotopically uniform pool reflecting  $\text{NH}_3$  fertilizer and/or soybean residue N to one that reflects partial nitrification of recently applied  $\text{NH}_3$  (Fig. 7; Table 1). This revealed variation in  $\delta^{15}\text{N}_{\text{source}}$  prompts two related questions: (1) how does the soil-plant system store, transport, and partition fertilizer- and/or soybean-derived N, making this N a potentially persistent source of tile drainage  $\text{NO}_3^-$  loss during unfertilized periods?; and (2) why is there a lack of isotopic signals of the bulk soil N pool and corn biomass in tile drainage  $\text{NO}_3^-$ ?

In our study system, if we assume a fertilizer use efficiency of 40%, which is at the upper end of reported range for cereal cropping (Yan et al., 2020), the amount of fertilizer N not recovered by the same-season corn was about  $120 \text{ kg N ha}^{-1} \text{ yr}^{-1}$ . Likewise, the amount of N input via incorporation of aboveground, non-grain soybean biomass was also on the order of  $120 \text{ kg N ha}^{-1} \text{ yr}^{-1}$ . Although a large fraction of this N input may be immobilized by soil microbes, the revealed pattern in  $\delta^{15}\text{N}_{\text{source}}$  dynamics seems to suggest a selective mechanism that preferentially stabilizes corn biomass N but renders fertilizer- and/or soybean residue-derived N a labile source for  $\text{NO}_3^-$  production and loss. This selective mechanism may be driven by stoichiometric imbalances and variable microbial N use efficiency at the soil microsite scale (Schimel and Bennett, 2004; Manzoni et al., 2008). For example, the higher C-to-N ratio of corn residues may create soil microsites with limited N availability, which may subsequently increase microbial N use efficiency and promote tighter recycling of microbial residues (Kaiser et al., 2014; Mooshammer et al., 2014; Zhang et al., 2019). This tightened recycling, combined with the slower decomposition of corn residues compared to soybean residues, may favor the stabilization and sequestration of soluble organic compounds in these N-poor microsites due to greater opportunities for protection through sorption toward mineral surfaces and incorporation into aggregates (Lehmann and Kleber, 2015; Cotrufo et al., 2015). On the other hand, in N-rich soil microsites, such as those receiving direct fertilizer input and/or soybean residues, high N availability may initially stimulate N immobilization and microbial growth (Mooshammer et al., 2014; Zhang et al., 2019). However, as this growth continues, microbial biomass may be increasingly carbon-limited,

which may eventually result in elevated microbial mortality, reduced microbial N use efficiency, and recycling of carbon stored in microbial residues (Cui et al., 2020; Li et al., 2021). The release of soluble organic compounds from this recycling may saturate the protection sinks of these N-rich microsites (Castellano et al., 2015), resulting in leakage of  $\text{NO}_3^-$  via mineralization and subsequent nitrification.

Additionally, the revealed chemostatic  $\text{NO}_3^-$  export regime and  $\delta^{15}\text{N}_{\text{source}}$  dynamics may also result from the large capacity of the soil profile to store water and  $\text{NO}_3^-$ . Using chloride as a conservative tracer, a previous study at the same field has revealed a total storage capacity of at least 1000 mm for tile flow generation (Yu et al. 2023). Assuming a porosity of 0.4 for a silt loam Mollisol, this storage would require a soil depth of at least 2.5 m, which corresponds to the entire above-tile profile plus an additional 1 m of soil below the tile drain. This extensive storage capacity may provide a significant mixing volume for  $\text{NO}_3^-$  that retains memories of past N input to the soil-plant system (Woo and Kumar, 2019; Williams et al., 2021). Importantly, the presence of this large mixing volume is consistent with the highly dampened variation in the  $\delta^{18}\text{O}_{\text{H}_2\text{O}}$  values of tile drainage compared to those of precipitation (Fig. 4e). However, on the other hand, a large storage and mixing volume of  $\text{NO}_3^-$  appears to be at odds with the high sensitivity of tile drainage  $\delta^{15}\text{N}_{\text{NO}_3}$  and  $\delta^{18}\text{O}_{\text{NO}_3}$  to high flow events (Fig. 5 middle and right columns) and with the direct response of  $\delta^{15}\text{N}_{\text{NO}_3}$  to fertilizer applications (Fig. 2c). One way to reconcile these seemingly paradoxical observations is the progressive activation of preferential flows as the soil becomes wetter (Klaus et al., 2013; Williams et al., 2016; Yu et al., 2023). In fine-textured soils with high root density, preferential flows may be facilitated by an extensive network of soil macropores associated with root channels, earthworm burrows, and desiccation cracks (Klaus et al., 2013; Williams et al., 2016). As  $\text{NO}_3^-$  in the upper depths of the soil profile may have lower  $\delta^{15}\text{N}_{\text{NO}_3}$  and  $\delta^{18}\text{O}_{\text{NO}_3}$  values due to surface inputs and higher mineralization and nitrification rates relative to denitrification, accelerated transport of water and  $\text{NO}_3^-$  from these near-surface layers via activated preferential flowpaths under wet conditions can explain the negative relationships between tile discharge and  $\text{NO}_3^-$  isotopes (Fig. 5 middle and right columns) and the flashy responses of tile drainage water and  $\text{NO}_3^-$  isotopes observed at the event scale

(Fig. 4c, 4d, and 4e) (Yu et al., 2023). Although  $\text{NO}_3^-$  concentration may also be higher at surface soil depths, the effect of this preferential transport was not readily detected by  $\text{NO}_3^-$  concentration measurements due to the simultaneous dilution of  $\text{NO}_3^-$  under high flow conditions.

We speculate that the combination of these proposed mechanisms based on soil-microbe interactions at the soil microsite scale and the storage and preferential transport of  $\text{NO}_3^-$  in the soil profile may account for the variability of  $\delta^{15}\text{N}_{\text{source}}$  in tile drainage. Although we are not able to verify these mechanisms and tease apart their relative importance due to the lack of appropriate data, the observed temporal variations in  $\text{NO}_3^-$  isotopes provide hints at the emergent modes of  $\text{NO}_3^-$  source and transport dynamics across the cropping phases. For example, the coupled increase in tile drainage  $\text{NO}_3^-$  concentrations and the decrease in  $\delta^{15}\text{N}_{\text{NO}_3}$  values observed during the corn non-growing season in 2021 and 2023 (Fig. 4b and 4c), combined with the large input of soybean residues during this period, may indicate that mineralization and recycling of soybean residue N were an important contributor of  $\text{NO}_3^-$  loss during this period. On the other hand, the similarity in  $\delta^{15}\text{N}_{\text{NO}_3}$  and  $\delta^{18}\text{O}_{\text{NO}_3}$  values observed during the soybean year to those measured at the end of the corn growing season in 2021 (Fig. 4b and 4c) may imply a delayed release of fertilizer-derived  $\text{NO}_3^-$  due to water storage and the early cessation of tile flows in the 2021 corn growing season.

To reduce  $\text{NO}_3^-$  loss from tile-drained midwestern agroecosystems, fertilizer management that aims to synchronize N input and crop N demand has been an organizing principle for improving N use efficiency in these systems. However, a recent synthesis based on 1000 site-years of tile drainage  $\text{NO}_3^-$  loads revealed no significant differences between different fertilizer timing and application methods in reducing tile  $\text{NO}_3^-$  loss (Christianson and Harmel, 2015). Based on the results of this study, we show that  $\text{NO}_3^-$  export in tile drainage may be modulated by a N legacy effect that can mask short-term responses – and thus evaluation – of tile drainage  $\text{NO}_3^-$  loads to changes in management practices. However, we note that the term “legacy effect” should not be taken literally, as any  $\text{NO}_3^-$  contained in a water sample reflects a mixture of N with different origins and introduced to the system at different times. The legacy effect is a question of degree, and there is evidence that extensive system modifications (e.g., conversion from corn/soybean to perennial biofuel crops) can significantly



tighten the N cycle in tile-drained agroecosystems over a short time span (Smith et al., 2013). Nevertheless, the existence of these legacy effects calls for long-term and well controlled field studies, which remain an exception rather than the norm in tile drainage research (Christianson and Harmel, 2015; Gentry et al., 2023). It also reinforces the importance of coupling N and carbon cycles in conventionally managed row crop systems via diversification of N source input and plant types in reducing tile drainage  $\text{NO}_3^-$  loss (Drinkwater and Snapp, 2007).

## 5. Conclusions

In this study, multiyear measurements of tile drainage  $\text{NO}_3^-$  isotopes were used to examine the sources and processes controlling  $\text{NO}_3^-$  export from a tile-drained con-soybean field. These field isotope data were supplemented by characterization of the N isotopic composition of potential  $\text{NO}_3^-$  sources (i.e., fertilizer, soil nitrogen, and crop biomass) in the field and by the oxygen isotopic composition of  $\text{NO}_3^-$  produced by nitrification in soil incubations. To our knowledge, this study represents the first attempt to adopt such a fine-grained approach based on  $\text{NO}_3^-$  isotopes to examine  $\text{NO}_3^-$  source and transport mechanisms in terrestrial settings. The results from the soil incubation demonstrated that the  $\delta^{18}\text{O}$  of  $\text{NO}_3^-$  produced by soil nitrification ( $\delta^{18}\text{O}_{\text{NO}_3,\text{nit}}$ ) is determined by isotopic O exchange between  $\text{NO}_2^-$  and water and by kinetic isotope effects during O atom incorporation from  $\text{O}_2$  and water, consistent with previous studies using nitrifying pure cultures. However,  $\delta^{18}\text{O}_{\text{NO}_3,\text{nit}}$  was significantly different between fertilized and non-fertilized treatments, indicating that the variability of  $\delta^{18}\text{O}_{\text{NO}_3,\text{nit}}$  is controlled by nitrification rate through its control on  $\text{NO}_2^-$  accumulation and, consequently, the potential of  $\text{NO}_2^-$ -water exchange during the nitrification process.

While  $\text{NO}_3^-$  export in tile drainage was characterized by prominent chemostasis,  $\text{NO}_3^-$  isotopes in tile drainage were highly sensitive to variations in tile discharge and fertilizer input. Following fertilizer applications, the dramatic decline in tile drainage  $\delta^{15}\text{N}_{\text{NO}_3}$  values indicates rapid nitrification of applied  $\text{NH}_3$  fertilizer and the subsequent loss of the nitrified  $\text{NO}_3^-$  possibly via preferential flows. On the other hand, variations in tile drainage  $\delta^{15}\text{N}_{\text{NO}_3}$  and  $\delta^{18}\text{O}_{\text{NO}_3}$  values were strongly coupled during unfertilized periods (i.e., corn non-growing season and soybean year), suggesting denitrification as a dominant process regulating the temporal variation of  $\text{NO}_3^-$  isotopes



during these periods. Moreover, this strong coupling provides evidence for an approximately constant  $\text{NO}_3^-$  isotopic source signature, as opposed to a dynamic mixing of multiple  $\text{NO}_3^-$  sources with distinct isotopic signatures. Combining the field  $\text{NO}_3^-$  and water isotope measurements with the O isotopic systematics of soil nitrification revealed a dynamic  $\text{NO}_3^-$  source mechanism, with fertilizer applications triggering a pulsed release of  $\text{NO}_3^-$  that shifts the isotopic signature of  $\text{NO}_3^-$  from an isotopically uniform pool reflecting  $\text{NH}_3$  fertilizer and/or soybean residue N to one that reflects partial nitrification of recently applied  $\text{NH}_3$ . These results imply that fertilizer- and/or soybean residue-derived N was a persistent source of tile drainage  $\text{NO}_3^-$  export during unfertilized periods and highlights a N legacy effect possibly resulting from N recycling at the soil microsite scale and a large storage and mixing capacity of  $\text{NO}_3^-$  in the soil. While more research is needed to examine the coupled biogeochemical and hydrological processes regulating the formation of legacy N stores and their impact on  $\text{NO}_3^-$  export in tile-drained agroecosystems, through this study, we demonstrated the unique strength of  $\text{NO}_3^-$  isotopes as an integrative tracer for probing complex N source and transport mechanisms under heterogeneous field conditions and how a rigorous understanding of isotopic fractionations (e.g., those underlying  $\delta^{18}\text{O}_{\text{NO}_3, \text{nit}}$ ) is key to maximizing this strength.

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## **Data Availability Statement**

The data that support the findings of this study, including those collected during the laboratory incubation and field tile drainage measurements, are available at Hu and Yu (2024).

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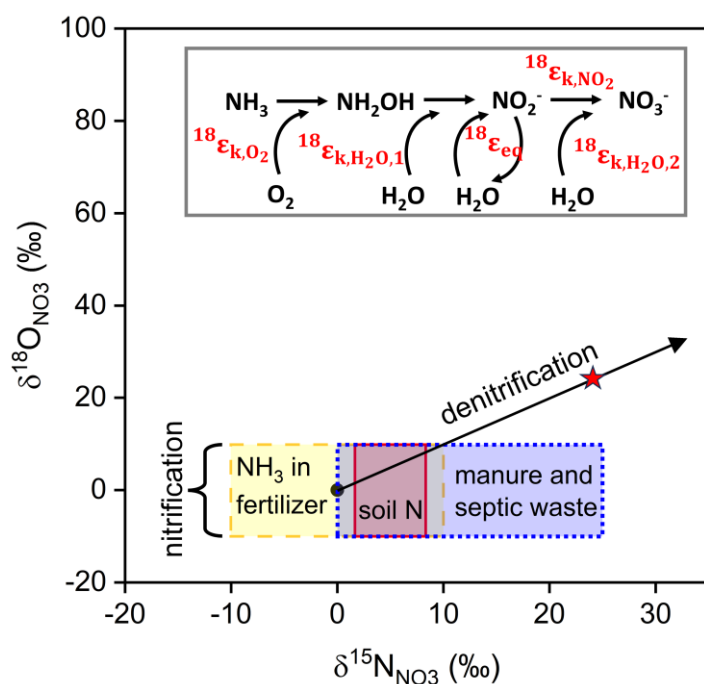
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## Table and Figures

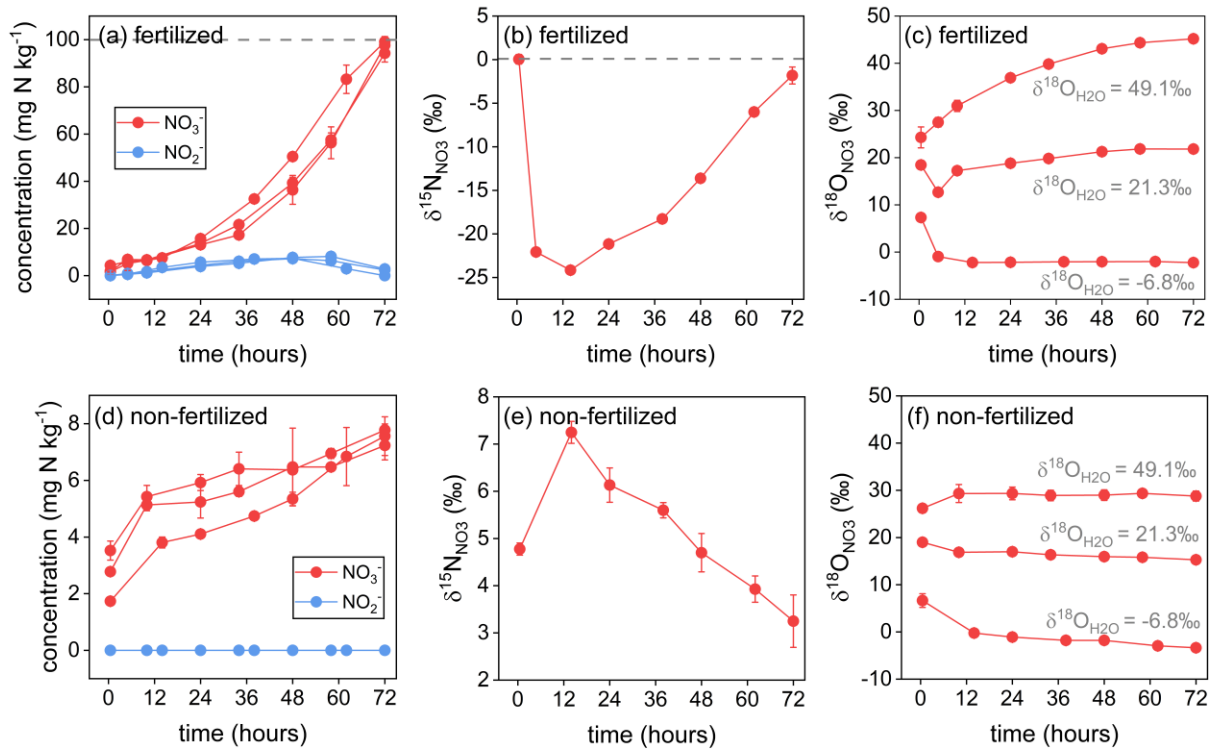
**Table 1.**  $\delta^{15}\text{N}_{\text{source}}$  values in ‰ for the three tiles and three cropping phases estimated based on  $\text{NO}_3^-$  load-weighted mean  $\delta^{18}\text{O}_{\text{H}_2\text{O}}$  values of tile drainage.

	Corn growing season		Corn non-growing season		Soybean year	
	$\delta^{18}\text{O}_{\text{H}_2\text{O}}$	$\delta^{15}\text{N}_{\text{source}}$	$\delta^{18}\text{O}_{\text{H}_2\text{O}}$	$\delta^{15}\text{N}_{\text{source}}$	$\delta^{18}\text{O}_{\text{H}_2\text{O}}$	$\delta^{15}\text{N}_{\text{source}}$
<b>Tile A</b>	-6.1	-7.1 to -3.0	-6.5	-3.1 to 0.8	-6.2	-2.6 to 1.4
<b>Tile B</b>	-5.9	-8.3 to -4.2	-6.4	-4.4 to -0.4	-6.2	-4.7 to -0.7
<b>Tile C</b>	-6.1	-8.0 to -4.0	-5.9	-3.8 to 0.3	-6.5	-4.8 to -0.9

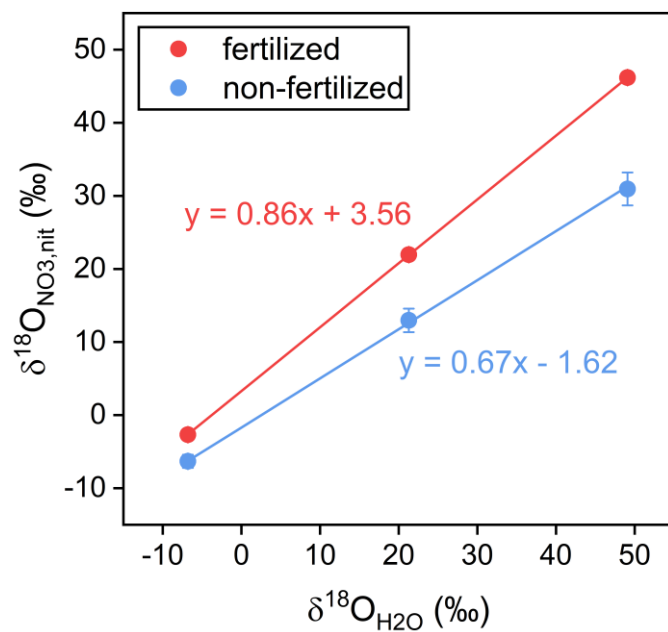


**Figure 1.** Dual  $\text{NO}_3^-$  isotope space adapted from Kendall et al. (2007). Shaded boxes indicate typical  $\delta^{15}\text{N}$  and  $\delta^{18}\text{O}$  values of  $\text{NO}_3^-$  derived or nitrified from various N sources. The typically assumed range of  $\delta^{18}\text{O}_{\text{NO}_3}$  values produced from nitrification of  $\text{NH}_3$  and organic N (-10 to 10‰) are denoted by “nitrification”. The black arrow indicates the linear denitrification trajectory that reflects the coupled N and O isotopic fractionations during denitrification. Along this arrow, the filled circle represents the initial isotopic composition (i.e., source signature) of  $\text{NO}_3^-$ , while the red star indicates  $\text{NO}_3^-$  isotopic composition after denitrification enrichment. A schematic diagram of O isotopic fractionations and exchange during nitrification is shown in the insert. Sources of O atoms ( $\text{O}_2$  and water) for  $\text{NH}_3$  and  $\text{NO}_2^-$  oxidation are shown, as well as O isotope effects associated with O atom incorporation ( $^{18}\epsilon_{\text{k},\text{O}_2}$ ,  $^{18}\epsilon_{\text{k},\text{H}_2\text{O},1}$ , and  $^{18}\epsilon_{\text{k},\text{H}_2\text{O},2}$ ),  $\text{NO}_2^-$ -water equilibrium ( $^{18}\epsilon_{\text{eq}}$ ), and  $\text{NO}_2^-$  oxidation ( $^{18}\epsilon_{\text{k},\text{NO}_2}$ ). Detailed descriptions of the O isotope effects are provided in Section 3.4.





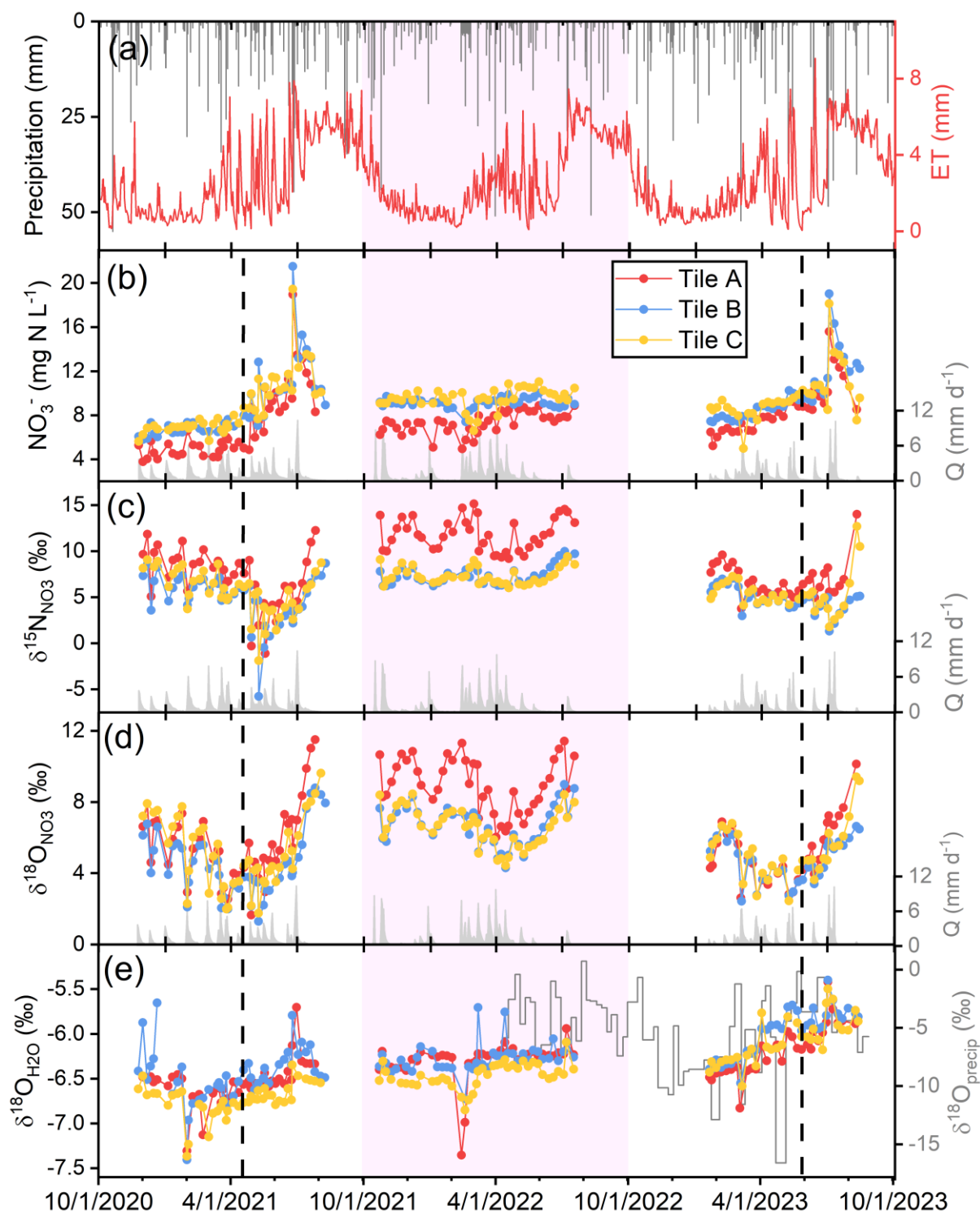
**Figure 2.** Temporal evolution of soil NO<sub>3</sub><sup>-</sup> and NO<sub>2</sub><sup>-</sup> concentrations (a and d), δ<sup>15</sup>N<sub>NO3</sub> values (b and e) and δ<sup>18</sup>O<sub>NO3</sub> values (c and f) for the three δ<sup>18</sup>O<sub>H2O</sub> treatments with and without fertilization. Error bars correspond to the standard deviation of replicate measurements. The horizontal dashed line denotes the initial concentration of amended NH<sub>4</sub><sup>+</sup> in (a) and the initial δ<sup>15</sup>N of amended NH<sub>4</sub><sup>+</sup> in (b). Text labels in (c) and (f) differentiate measured δ<sup>18</sup>O<sub>NO3</sub> values among the three δ<sup>18</sup>O<sub>H2O</sub> treatments.



**Figure 3.** Dependence of  $\delta^{18}\text{O}_{\text{NO}_3,\text{nit}}$  on  $\delta^{18}\text{O}_{\text{H}_2\text{O}}$  under the fertilized and non-fertilized conditions.

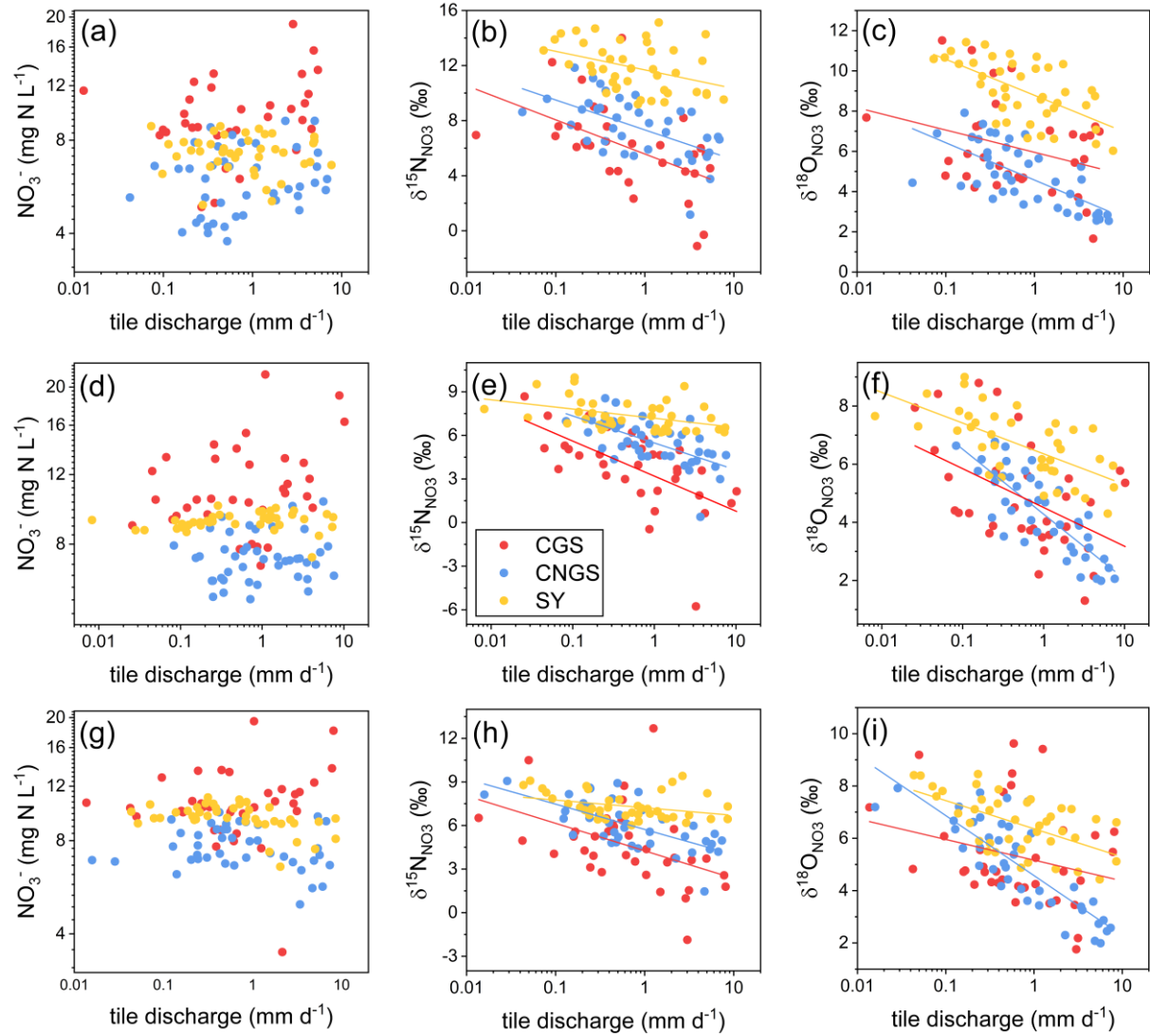
Equations correspond to the linear regression fit of the observations.



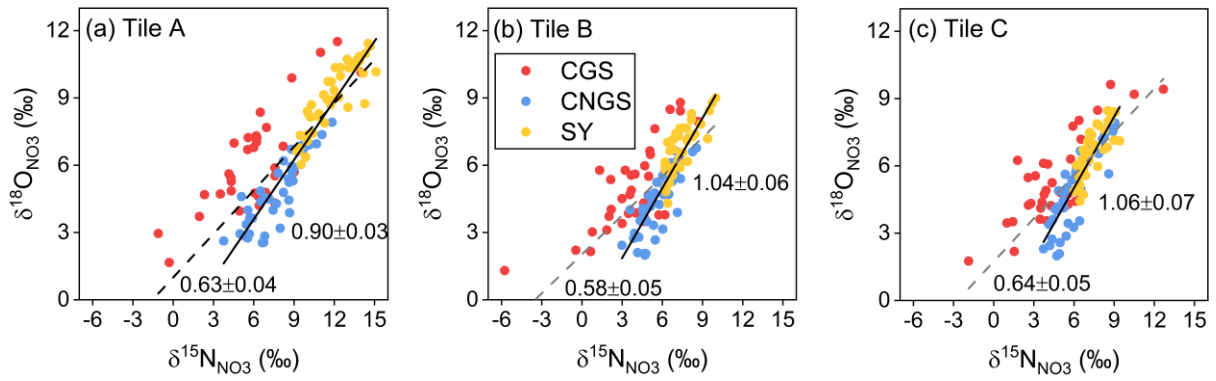


**Figure 4.** Temporal variations of precipitation and evapotranspiration (ET) (a),  $\text{NO}_3^-$  concentrations in tile drainage (b),  $\delta^{15}\text{N}_{\text{NO}_3}$  values (c),  $\delta^{18}\text{O}_{\text{NO}_3}$  values (d), and  $\delta^{18}\text{O}_{\text{H}_2\text{O}}$  values of tile drainage and weekly cumulative precipitation (e) at the tile-drained field. The black vertical dashed lines indicate the timing of anhydrous  $\text{NH}_3$  application in the 2021 and 2023 corn years, respectively. Faded area

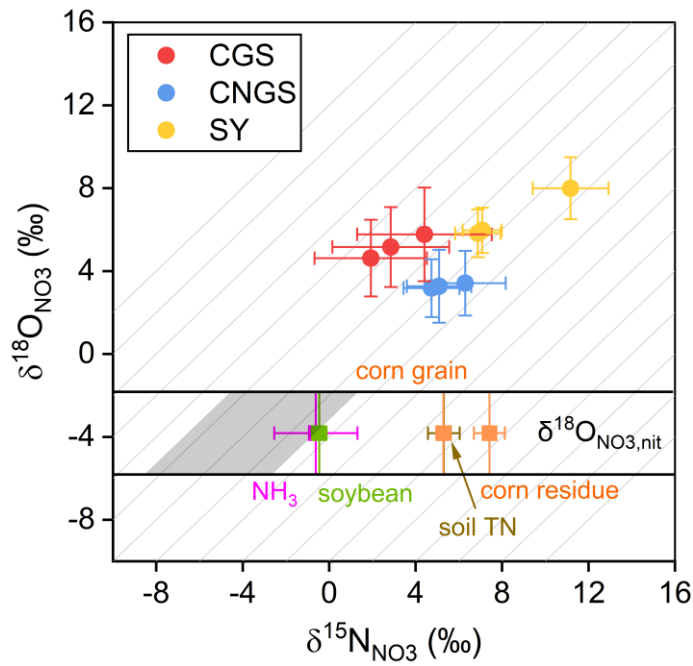
denotes the 2022 soybean year. Tile discharge (Q) measured at Tile B is shown in gray in panels b, c, and d to aid interpretation of  $\text{NO}_3^-$  concentration and isotope data.



**Figure 5.** Relationships between tile discharge and  $\text{NO}_3^-$  concentration (a, d, g; left column), between tile discharge and  $\delta^{15}\text{N}_{\text{NO}_3}$  value (b, e, h; middle column), and between tile discharge and  $\delta^{18}\text{O}_{\text{NO}_3}$  value (c, f, i; right column), for the three tiles (top row for Tile A, middle row for Tile B, and bottom row for Tile C) and three cropping phases. Note that  $\text{NO}_3^-$  concentration and tile discharge are shown in logarithmic scale. Linear regression fits for log-transformed tile discharge and  $\text{NO}_3^-$  isotopes significant at the 0.05 level are also shown. ‘CGS’, ‘CNGS’, and ‘SY’ in the legend denote corn growing season, corn non-growing season, and soybean year, respectively.



**Figure 6.** Biplots of  $\delta^{15}\text{N}_{\text{NO}_3}$  versus  $\delta^{18}\text{O}_{\text{NO}_3}$  values for the three tiles and three cropping phases (a-c). The solid line denotes linear regression based on  $\text{NO}_3^-$  isotopes measured during the corn non-growing season and soybean season. The dashed line denotes linear regression based on all  $\text{NO}_3^-$  isotope data. Linear regression slopes ( $\pm 1$  SE) are also shown. ‘CGS’, ‘CNGS’, and ‘SY’ in the legend denote corn growing season, corn non-growing season, and soybean year, respectively.



**Figure 7.**  $\text{NO}_3^-$  source characterization in dual isotope space. Area encompassed by the solid horizontal lines denotes the predicted range of  $\delta^{18}\text{O}_{\text{NO}_3,\text{nit}}$  based on the empirical  $\delta^{18}\text{O}_{\text{NO}_3,\text{nit}}$ -versus- $\delta^{18}\text{O}_{\text{H}_2\text{O}}$  relationships derived from the soil nitrification experiment. Squares and error bars correspond to the  $\delta^{15}\text{N}$  and  $\delta^{18}\text{O}$  values of  $\text{NO}_3^-$  nitrified from applied  $\text{NH}_3$  fertilizer (magenta), soil total N (brown), soybean biomass N (green), and corn biomass N (orange). Dots denote  $\text{NO}_3^-$  load-weighted mean  $\delta^{15}\text{N}_{\text{NO}_3}$  and  $\delta^{18}\text{O}_{\text{NO}_3}$  values for the three tiles and three cropping phases. The error bars of the dots reflect the standard deviation of weekly  $\delta^{15}\text{N}_{\text{NO}_3}$  and  $\delta^{18}\text{O}_{\text{NO}_3}$  values measured during each tile and cropping phase. The grey shaded area corresponds to the range of  $\delta^{15}\text{N}_{\text{source}}$  when considering all tiles and cropping phases collectively. Diagonal grey lines denote denitrification trajectories with a slope of 1. ‘CGS’, ‘CNGS’, and ‘SY’ in the legend denote corn growing season, corn non-growing season, and soybean year, respectively.