

Development of a deep-water carbonate ion concentration proxy based on preservation of planktonic foraminifera shells quantified by X-ray CT scanning

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December 16, 2022

Abstract

The quantitative and objective characterization of dissolution intensity in fossil planktonic foraminiferal shells could be used to reconstruct past changes in bottom water carbonate ion concentration. Among proxies measuring the degree of dissolution of planktonic foraminiferal shells, X-ray micro-Computed Tomography (CT) based characterization of apparent shell density appears to have good potential to facilitate quantitative reconstruction of carbonate chemistry. However, unlike the well-established benthic foraminiferal B/Ca ratio-based proxy, only a regional calibration of the CT-based proxy exists based on a limited number of data points covering mainly low-saturation state waters. Here we determined by CT-based proxy the shell dissolution intensity of planktonic foraminifera *Globigerina bulloides*, *Globorotalia inflata*, *Globigerinoides ruber*, and *Trilobatus sacculifer* from a collection of core top samples in the Southern Atlantic covering higher saturation states, and assessed the characteristics and reliability of CT-based proxy. We observed that the CT-based proxy is generally controlled by deep-water $\Delta[\text{CO}_3^{2-}]$ like the B/ δ proxy, but its ε varies from -20 to $10 \mu\text{mol kg}^{-1}$. In this range, the CT-based proxy appears directly and strongly related to deep-water $\Delta[\text{CO}_3^{2-}]$, whereas the B/ δ of benthic foraminifera appears to be affected by porewater saturation in carbonate-rich substrates. On the other hand, the $\delta^{18}\text{O}$ -based proxy is affected by diagenetic dissolution in areas with high productivity. Like the B/ δ proxy, the $\delta^{18}\text{O}$ -based proxy requires species-specific calibration, but the ε varies among species in a different way in the proxy.

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5

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17 **Key Points:**

- 18 • Planktonic foraminiferal shell dissolution intensity was determined by CT-based proxy
19 using core top samples in the Southern Atlantic Ocean.
- 20 • The characteristics and reliability of CT-based proxy were assessed by comparing with
21 conventional proxy.
- 22 • The CT-based proxy is available for reconstructing deep seawater carbonate ion
23 concentration under the appropriate condition.
24
25

26 **Abstract**

27 The quantitative and objective characterization of dissolution intensity in fossil planktonic
28 foraminiferal shells could be used to reconstruct past changes in bottom water carbonate ion
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32 chemistry. However, unlike the well-established benthic foraminiferal B/Ca ratio-based proxy,
33 only a regional calibration of the CT-based proxy exists based on a limited number of data points
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35 dissolution intensity of planktonic foraminifera *Globigerina bulloides*, *Globorotalia inflata*,
36 *Globigerinoides ruber*, and *Trilobatus sacculifer* from a collection of core top samples in the
37 Southern Atlantic covering higher saturation states, and assessed the characteristics and
38 reliability of CT-based proxy. We observed that the CT-based proxy is generally controlled by
39 deep-water $\Delta[\text{CO}_3^{2-}]$ like the B/Ca proxy, but its effective range of $\Delta[\text{CO}_3^{2-}]$ is between -20 to
40 $10 \mu\text{mol kg}^{-1}$. In this range, the CT-based proxy appears directly and strongly related to deep-
41 water $\Delta[\text{CO}_3^{2-}]$, whereas the B/Ca of benthic foraminifera appears to be affected by porewater
42 saturation in carbonate-rich substrates. On the other hand, the CT-based proxy is affected by
43 supralysoclinal dissolution in areas with high productivity. Like the B/Ca proxy, the CT-based
44 proxy requires species-specific calibration, but the effect of species-specific shell difference in
45 susceptibility to dissolution on the proxy is small.

46

47 **1 Introduction**

48 The atmospheric CO_2 concentration has fluctuated by ~ 80 ppm between glacial and
49 interglacial periods, implying a large and rapid exchange of carbon between the atmosphere and
50 the ocean on those time scales [Barnola et al., 1987; Petit et al., 1999]. This is because the
51 oceanic carbon pool is about 50 times larger than that of the atmosphere [Sigman and Boyle,
52 2000], and carbon storage in the deep ocean and changes in deep-water circulation can
53 substantially alter atmospheric $p\text{CO}_2$. However, data concerning the amount of carbon storage in
54 the deep sea and their temporal and spatial variation, which are essential to understand the
55 glacial-interglacial $p\text{CO}_2$ exchange between ocean and atmosphere, is insufficient. Because the
56 deep seawater carbonate ion concentration ($[\text{CO}_3^{2-}]$) is governed primarily by the concentration
57 of dissolved inorganic carbon and alkalinity, its reconstruction can provide valuable insights into
58 the changes in the global carbon cycle.

59 The B/Ca ratio of epifaunal benthic foraminifera has been proposed as a quantitative
60 deep seawater $[\text{CO}_3^{2-}]$ proxy [e.g., Yu and Elderfield, 2007]. This proxy relies on the fact that
61 the ratio of the two major boron species in the deep seawater, $\text{B}(\text{OH})_3$ and $\text{B}(\text{OH})_4^-$, varies with
62 pH [Hemming et al., 1992] and these variations are reflected in the B/Ca ratio in shell calcite.
63 Indeed, a significant correlation between B/Ca ratio in the shells of *Cibicidoides wuellerstorfi*
64 and deep seawater $\Delta[\text{CO}_3^{2-}]$ was shown by Yu and Elderfield [2007]. The sensitivity of *C.*
65 *wuellerstorfi* B/Ca ratio to deep seawater $\Delta[\text{CO}_3^{2-}]$ was evaluated based on core-top calibration
66 at more than 200 sites, indicating an uncertainty of $\pm 5 \mu\text{mol kg}^{-1}$ in $[\text{CO}_3^{2-}]$ (Brown and
67 Elderfield, 1996, Rae et al., 2011, Raitzsch et al., 2011, Yu and Elderfield., 2007, Yu et al., 2013,
68 2014, Brown et al., 2011). Based on this calibration, subsequent studies reconstructed temporal
69 variations in deep seawater $[\text{CO}_3^{2-}]$ at several sites, where the sediment samples contain

70 sufficient amount of benthic foraminifera shells of the target species [e.g., Yu et al., 2016, 2020,
71 Allen et al., 2015, 2019]. Next to the most frequently used calibration based on *C. wuellerstorfi*,
72 there also exist data for another epifaunal species, *C. mundulus*, confirming a relationship
73 between shell B/Ca and deep seawater $\Delta[\text{CO}_3^{2-}]$, but the calibration shows a different slope,
74 indicating that the incorporation of B into the shell calcite is affected by species-specific
75 processes. Therefore, the B/Ca method can only be applied where the same species occurs
76 throughout the studied interval and also where a sufficient number of shells can be recovered to
77 facilitate the chemical analysis. This prevents applications in settings where the regional
78 environmental change caused large shifts in food availability and bottom water oxygen content,
79 resulting in shifts in benthic communities, as well as in settings with high sedimentation rate,
80 where the concentration of foraminifera shells is low (e.g. Kitazato et al., 2000, Gooday, 2003,
81 Geslin et al., 2004).

82 Planktonic foraminifera shells are a primary component of carbonate in the deep-sea
83 sediment (Schiebel, 2002), and the preservation of these shells is intimately associated with the
84 deep seawater $\Delta[\text{CO}_3^{2-}]$ (e.g., Berger et al., 1982). Therefore, another way to reconstruct deep
85 seawater $[\text{CO}_3^{2-}]$ is by the quantification of the degree of dissolution of planktonic foraminifera
86 shells (e.g. Berger et al., 1982). Previous studies made use of this relationship, but applied
87 different ways to quantify dissolution intensity, such as by shell fragmentation, proportion of
88 dissolution resistant species or the ratio between benthic and planktonic foraminifera (Berger et
89 al., 1982, Peterson and Prell, 1985, Kucera, 2007). These proxies all show the expected direction
90 of the relationship with deep seawater $\Delta[\text{CO}_3^{2-}]$, but are either hard to objectively quantify or are
91 affected by initial conditions (assemblage composition) (Kucera, 2007). The size-normalized
92 weight of sedimentary shell was proposed to quantify carbonate dissolution intensity more
93 objectively (Lohmann, 1995, Broecker and Clark, 2001). However, the initial weight of the
94 planktonic foraminifera shell is also controlled by surface water properties, such as surface water
95 carbonate chemistry (e.g., Barker and Elderfield, 2002; Marshall et al., 2013) making
96 quantitative reconstructions of deep-water $[\text{CO}_3^{2-}]$ impossible in settings where surface
97 properties have changed as well.

98 More recent studies suggested that the problem of quantifying preservation state of
99 planktonic foraminifera shells can be circumvented by measurements of shell architecture using
100 X-ray micro-CT scanning (Johnstone et al., 2010, Iwasaki et al., 2015, Iwasaki et al., 2019). In
101 particular, the quantification of shell preservation by measuring the proportion of more strongly
102 dissolved calcite, identified by their lower CT number, provides a means to objectively
103 determine the preservation state in a way that is not dependent on the initial size or weight of the
104 shell. As long as the ratio of the different types of calcites, making up the lamellar structure of
105 the shells remains the same, the proportion of the more dissolved calcite should only be related
106 to deep-water $[\text{CO}_3^{2-}]$ (Iwasaki et al., 2015, Iwasaki et al., 2019). However, so far only a regional
107 South Pacific calibration of the CT-based proxy exists (Iwasaki et al., 2022). This calibration is
108 based on a limited number of data points, covering mainly low saturation state waters, and it has
109 not been directly compared with the B/Ca proxy.

110 In addition, there are factors other than deep seawater $[\text{CO}_3^{2-}]$ variation that control
111 carbonate dissolution on the deep seafloor and which may affect the CT-based proxy differently
112 than the B/Ca proxy. The first factor is the variation in sedimentation rate, which alters the
113 exposure time to deep seawater and results in differences in dissolution intensity of planktonic
114 foraminifera shells (Berger et al., 1982). The second factor is the decomposition of organic
115 material in the sediment, which decreases ambient seawater (or porewater) $[\text{CO}_3^{2-}]$ and may

116 cause carbonate dissolution in seafloor sediments deposited above the regional lysocline (Berger
117 et al., 1970, Milliman, 1993, Hales and Emerson, 1996, Hales, 2003). Finally, it is known from
118 earlier attempts to use the preservation of planktonic foraminifera shells as a proxy for deep-
119 water properties that significant changes in preservation are only encountered below a certain
120 critical threshold value of deep seawater $[\text{CO}_3^{2-}]$, and shells deposited above this foraminiferal
121 lysocline all appear well preserved (Berger et al., 1982). Therefore, to develop the method using
122 CT-scanning of planktonic foraminiferal shells as paleo-deep seawater $[\text{CO}_3^{2-}]$ proxy, assessing
123 the effect of these secondary factors on the preservation state of shells deposited in various
124 sedimentary settings is necessary.

125 To assess the accuracy and reliability of the CT-based proxy for deep seawater $[\text{CO}_3^{2-}]$
126 reconstruction, we have quantified planktonic foraminifera shell preservation in a collection of
127 samples from the South Atlantic, covering a gradient towards higher ambient $\Delta[\text{CO}_3^{2-}]$ values,
128 and representing different sedimentation rates and productivity regimes. Importantly, in all the
129 analyzed samples, B/Ca of *C. wuellerstorfi* have been determined by (Rae et al., 2011, Raitzsch
130 et al., 2013), allowing direct proxy intercomparison. We quantified the degree of shell
131 dissolution in four planktonic foraminifera species (*G. bulloides*, *G. inflata*, *G. ruber*, and *T.*
132 *sacculifer*) and propose a new way to quantify the proportion of more dissolved calcite, which
133 does not require instrument-specific calibration of the strength of the X-ray beam. We combine
134 the new measurements with previously obtained CT images (Iwasaki et al., 2019, 2022) to
135 determine the regional and species-specific effects and establish a new calibration formula,
136 allowing reconstructions of deep seawater $\Delta[\text{CO}_3^{2-}]$.

137

138 2 Materials and Methods

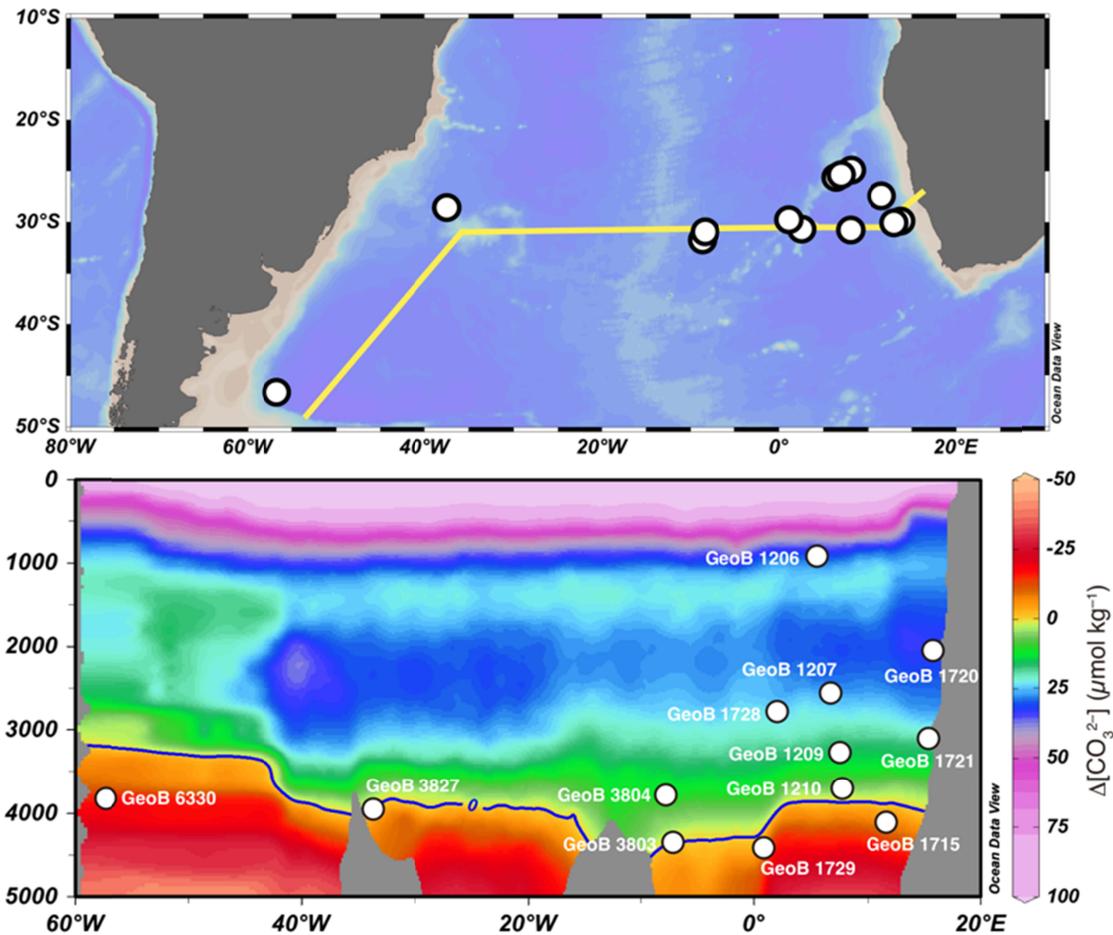
139 2.1. Multiple core samples and physicochemical properties

140 In this study, we used planktonic foraminifera shells extracted from 13 core top samples
141 collected by multiple corer between 940 and 4173 m water depth in the South Atlantic (Figure
142 1). The samples represent a subset of core tops where benthic foraminiferal (*Cibicidoides*
143 *wuellerstorfi*) B/Ca ratio, a conventional proxy of deep seawater $[\text{CO}_3^{2-}]$, has been determined
144 by previous studies (Rae et al., 2011, Raitzsch et al., 2013) (Figure 2). The subset contains all
145 samples used in the previous studies for which the material is stored in the GeoB repository in
146 Bremen and where information is available on sedimentation rate and accumulation rate of
147 organic carbon. From each core top sample, eight shells of each of four planktic foraminiferal
148 species (*G. bulloides*, *G. inflata*, *G. ruber*, and *T. sacculifer*) were collected for analysis by X-ray
149 micro-CT scanning. Among the 13 sites, shells of *G. bulloides* could be collected from all sites,
150 shells of *G. inflata* and *T. sacculifer* were collected from 12 sites, and shells of *G. ruber* were
151 collected from 11 sites. The all shells were collected from the 300-355 μm size fraction, and are
152 not fragmented or containing fractures and holes, or peeling of the surface. Estimates of
153 sedimentation rate (cm kyr^{-1}) at the studied sites were taken from age models of sediment cores
154 sampled at the same site (Wefer et al., 1990, Schulz et al., 1992, 1996; Bleil et al. 2001), and
155 used to calculate accumulation rate of organic carbon ($\text{g cm}^{-2} \text{kyr}^{-1}$) making use of total organic
156 carbon content (%TOC) and dry bulk density (g cm^{-3}) data for each site (Mollenhauer et al.,
157 2004).

158 In addition to the above sample set, we used a subset of published 16 core top samples
159 from the Pacific Ocean to extend the dissolution intensity calibration over a larger range of deep

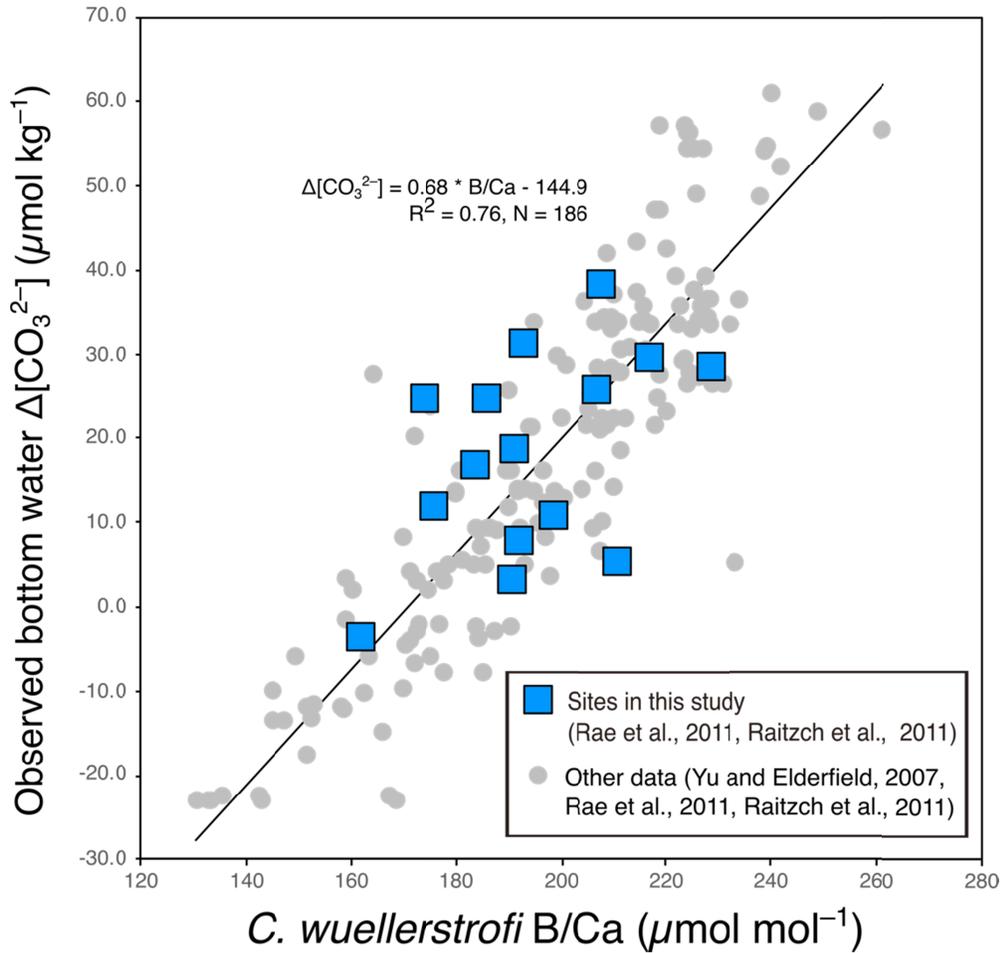
160 seawater $\Delta[\text{CO}_3^{2-}]$ values (Iwasaki et al., 2019, 2022). The core top sediment samples were
 161 obtained by multiple core sampling from the water depth between 1500 and 4000 m, with most
 162 of them located below the dissolution transition depth level ($\Delta[\text{CO}_3^{2-}] < 10 \mu\text{mol kg}^{-1}$). From
 163 each core top sample, more than eight shells of each of three planktic foraminiferal species (*G.*
 164 *bulloides*, *G. ruber*, and *T. sacculifer*) without external damage were collected from 200-355 μm
 165 size fraction for analysis by X-ray micro-CT scanning in previous studies, and the existing scans
 166 were here used for the calculation of the newly defined CT based dissolution index (Iwasaki et
 167 al., 2019 and 2022). To remain consistent with the existing data and due to the difficulty to
 168 unambiguously and consistently separating both species (Morard et al., 2019), specimens
 169 designated as *G. ruber* may contain both *G. ruber albus* and *G. elongatus*.

170 Bottom water values of $\Delta[\text{CO}_3^{2-}]$ for the all of studied sites were calculated from
 171 physicochemical parameters (temperature, salinity, total alkalinity, total inorganic carbon, and
 172 concentrations of phosphate and silicate) measured at nearby Global Ocean Data Analysis
 173 Project (GLODAP) profiles (Key et al., 2004). The site information and the estimated values of
 174 bottom water physicochemical properties for all Atlantic and Pacific sites are summarized in
 175 **Table 1**.
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 178 Figure 1. The locations of multiple corer samples in the South Atlantic used in this study. The
 179 section along the yellow line is showing the spatial distribution of the degree of carbonate
 180 saturation ($\Delta[\text{CO}_3^{2-}]$) in the study area.
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Figure 2. B/Ca ratio in shells of the benthic foraminifera *Cibicidoides wuellerstorfi* in core top samples plotted against deep-water $\Delta[\text{CO}_3^{2-}]$. The data of B/Ca ratio of benthic foraminifera in the core-top sample used in this study are shown by blue squares. Data are derived from Rae et al. (2011), Raitzch et al. (2011), and Yu and Elderfield (2007).

Cruise	Site	Latitude (S)	longitude (W)	Depth (m)	Deep water $\Delta[\text{CO}_3^{2-}]$ ($\mu\text{mol kg}^{-1}$)	Linear Sedimentation Rate (cm kyr^{-1})	TOC accumulation rate ($\text{g m}^{-2} \text{y}^{-1}$)	Caronate content (%)	C. w_B/Ca ($\mu\text{mol mol}^{-1}$)	error
M 12/1	1206-1	24.67	-6.48	940	25.1	No data	No data	No data	193	4
M 12/1	1207-2	24.59	-6.85	2593	29.9	1.5	0.17	95.5	217	7
M 12/1	1209-1	24.50	-7.28	3303	14.6	3.3	0.41	95.9	191	9
M 12/1	1210-3	24.48	-7.43	3750	6.5	1.4	0.16	93.7	192	5
M 20/2	1715-1	26.47	-11.63	4097	-5.9	10.0	1.73	83.0	191	8
M 20/2	1720-3	29.00	-13.82	2004	33.2	6.0	3.30	78.8	208	9
M 20/2	1721-5	29.17	-13.08	3045	18.9	6.0	1.41	84.8	186	9
M 20/2	1728-3	29.83	-2.40	2887	23.8	2.0	0.30	94.9	229	7
M 20/2	1729-1	28.88	-1.00	4401	-0.1	0.6	0.08	93.6	211	6
M 34/3	3803-1	30.34	8.57	4173	3.1	0.8	0.16	90.0	199	10
M 34/3	3804-2	30.74	8.77	3882	7.8	0.7	0.12	92.2	184	10
M 34/3	3827-1	25.02	38.54	3842	-4.3	0.9	0.18	50.5	176	9
M 46/3	6330-1	46.15	57.56	3874	-13.1	No data	No data	No data	162	7

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Table 1: Multiple core sample locations, depth (m), Deep water degree of $\Delta[\text{CO}_3^{2-}]$ ($\mu\text{mol kg}^{-1}$), Linear Sedimentation Rate (cm kyr^{-1}), TOC (Total Organic Carbon) accumulation rate ($\text{g m}^{-2} \text{y}^{-1}$) and *Cibicidoides wuellerstorfi* B/Ca ratio ($\mu\text{mol mol}^{-1}$) at each sampling site.

2.2. X-ray Micro-CT Scanning

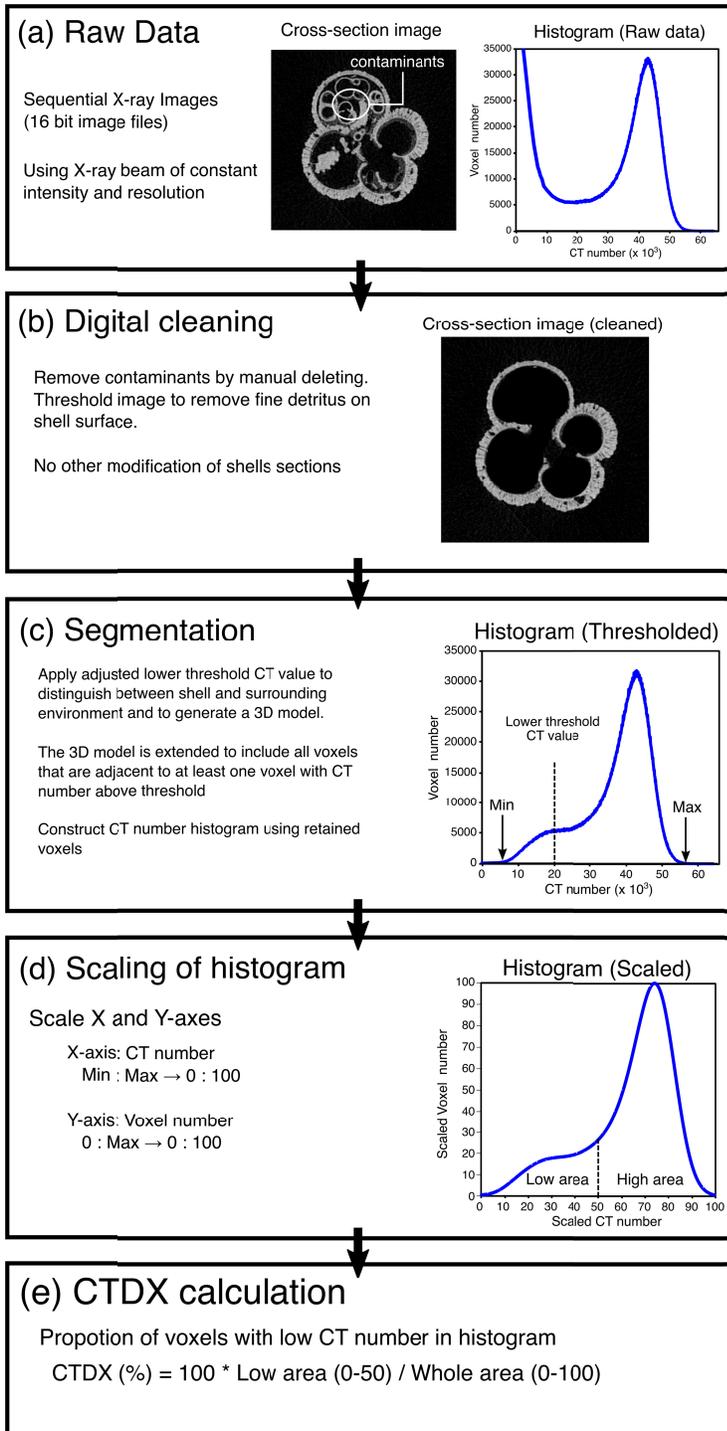
The XMCT system (ScanXmate-D160TSS105/11000, WhiteRabbit Corp., Tokyo, Japan) at the Japan Agency for Marine-Earth Science and Technology, Yokosuka, Japan, was used to obtain three-dimensional X-ray images of foraminiferal shells. The imaging was carried out with a high-resolution setting (X-ray focus spot diameter, 0.8 μm ; X-ray tube voltage, 80 kV; detector array size, 2000 \times 1336; 1500 projections/360°; 0.5 s/projection). After XMCT scanning, ConeCTexpress software (WhiteRabbit Corp., Tokyo, Japan) was used to convert the raw tomography data into segmented images of foraminifera shells. Image cross-sections were reconstructed from filtered back projections following the general principle of Feldkamp cone-beam reconstruction. The scanning and data processing methods followed a previous study [Iwasaki et al., 2015]. The CT number, indicating calcite density and visualizable porosity, was calculated based on the X-ray attenuation coefficient of each sample. We used Molcer Plus 3-D imaging software (WhiteRabbit Corp., Tokyo, Japan) to obtain isosurface images of the shells. Then, we evaluated the CT number histograms of each shell based on the 3-D tomography data.

2.3. Protocol of CT data processing for calculating CTDX

The CT scanning-based dissolution index (CTDX) of the planktic foraminiferal test employed in this study follows the same concept as the proxy of %Low-CT-number calcite volume previously suggested (Iwasaki et al., 2019, 2022). However, in this study, we modify the %Low-CT-number calcite volume proxy in a way that the resulting value is not dependent on the exact settings of the scanner, enabling applications using other X-ray micro-CT scanners, which had been modified to remove beam-hardening effects for assessing quantitative bulk density (i.e. submicron scale porosity) of organic carbonate. The protocol to calculate the value of CTDX from Serial Cross-Sectional Images of each specimen is shown in Figure 3. First, from the initial raw data (sequential sectional images) (Figure 3a), contaminants (e.g., detritus, fragments and other small foraminifera shells) stuck inside the test were removed by manual digital image processing (Figure 3b). After that, to segment shell material from other environments, we determined the lower threshold CT value by visual inspection of multiple specimens, ensuring only shell material is included. Under the process of segmentation, to facilitate a replicable smoothing of the shell surface and to make sure all voxels attached to the shell surface are included, we introduce a step where the six-neighboring voxels of a voxel located on the air-calcite boundary are retained. This is important to maintain voxels that partially contain shell material for the analysis. CT values in voxels retained after cleaning, segmentation, and thresholding were used to generate a CT histogram of the foraminiferal shell that could be used to calculate the dissolution index (Figure 3c). Previous studies [Iwasaki et al. 2019, 2022] suggested that the CT number histogram changes from monomodal to bimodal distribution with increasing dissolution. Based on this characteristic of change in the histogram shape with test dissolution, they proposed the relative volume of low-CT-number calcite to the volume of calcite in the whole shell (%Low-CT-number calcite volume) as a quantitative carbonate dissolution proxy for dissolution. Here we build on this approach but develop a new Dissolution Index (CTDX), which defines the threshold CT value independently of the scanner setting, using the shape of the CT value histogram. To this end, we first scaled the original CT number histogram by assigning 0 and 100 to the voxels with the lowest and highest CT values (Figure 3d). After that, we calculated the CTDX as the area ratio of the lower part of the histogram (X-axis: 0-50) to the whole histogram (X-axis: 0-100) (Figure 3e). Finally, we assigned a CTDX value to a sample as the average CTDX value of at least eight individual specimens. The minimum number

241 of shells needed to provide a representative estimate of the average preservation state of a given
 242 foraminifera species was determined by [Iwasaki et al. \(2022\)](#).

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 246 Figure 3. The protocol to calculate the foraminiferal shell dissolution index (CTDX) starts from
 247 raw sequential X-ray image data (a). (b) Contaminants in the shell sample are removed by
 248 manual deleting. (c) Apply threshold to distinguish between calcite material and surrounding

249 environment. Extract CT number histogram from cleaned and threshold 3D data. (d) Scaling X
250 and Y-axes of CT number histogram. (e) Calculation of CTDX (%).
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252 3 Results

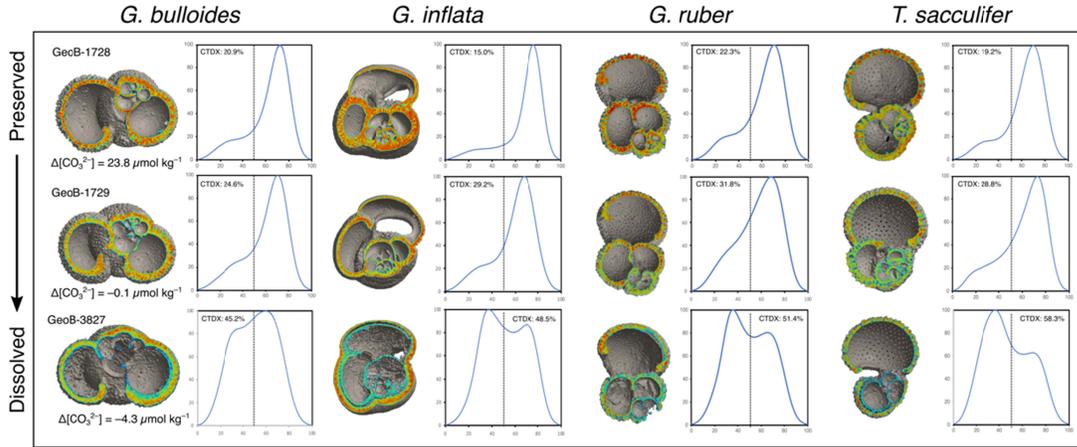
253 3.1. Evaluating test dissolution by X-ray micro-CT scanning

254 The X-ray micro-CT scanning enables us to observe the internal structure of the
255 foraminiferal shell and calcite density distribution in a single specimen's shell. The CT number
256 histogram provides an objective and quantitative means to evaluate microscale calcite density
257 distribution in each specimen. The **Figure 4** shows the changes in the internal shell structure and
258 CT number histogram of four species of planktic foraminifera in three seafloor sediment samples
259 representing different locations and bottom water $\Delta[\text{CO}_3^{2-}]$ conditions. As expected, and in
260 agreement with previous studies (Johnson et al., 2010, Iwasaki et al., 2019), we find a
261 progression of shell dissolution with decreasing $\Delta[\text{CO}_3^{2-}]$. The dissolution first affects the septa
262 among the juvenile chambers and then spreads to the inner layer of the chambers in the final
263 whorl. With progressive dissolution, the CT number histograms show a shift from a unimodal
264 distribution with a single, high CT number peak to a bimodal distribution, followed by a trend
265 toward a unimodal density distribution with a peak at a low CT number. These are consistent
266 with the results of previous studies (Iwasaki et al., 2015, 2019), suggesting that the shape of the
267 CT number histogram, irrespective of the CT scanning settings, can be used to describe the
268 preservation state of the foraminiferal shell.

269 Assuming that the shape of the CT number histogram, converted to the CTDX value as
270 described above, quantify the degree of shell alteration due to selective dissolution (removal) of
271 the most susceptible parts of the shell, the CTDX-based average state of shells of a given species
272 of planktonic foraminifera should show a systematic relationship with deep-water chemistry.
273 This is important, because the CTDX value is independent of the size and shape of the analyzed
274 individual shells. It describes the portion of the shell that is affected by dissolution, irrespective
275 of the analyzed calcite volume. In theory, the CTDX could even be applied to multiple species of
276 foraminifera, but because the shell architectures among species vary significantly, we began by
277 comparing the *G. bulloides* CTDX with that of three other species (*G. ruber*, *T. sacculifer*,
278 and *G. inflata*) in the same sediment samples to also evaluate the inter-species variation in
279 CTDX (**Figure 5**). The results shows that CTDX of *G. bulloides* significantly correlates with the
280 other three species (*G. ruber*: $R^2 = 0.7$, *T. sacculifer*: $R^2 = 0.72$, *G. inflata*: $R^2 = 0.81$), suggesting
281 that the CTDX of each of the species should be applicable as a dissolution proxy. However, the
282 CTDX values among the species are offset, in particular for the relationship between CTDX of
283 *G. bulloides*, *T. sacculifer* and *G. inflata*, the slopes of regression lines are relatively high (*T.*
284 *sacculifer*: 1.69, *G. inflata*: 1.39). This implies that *T. sacculifer* and *G. inflata* are more sensitive
285 to dissolution expressed by the CTDX index than *G. bulloides*.

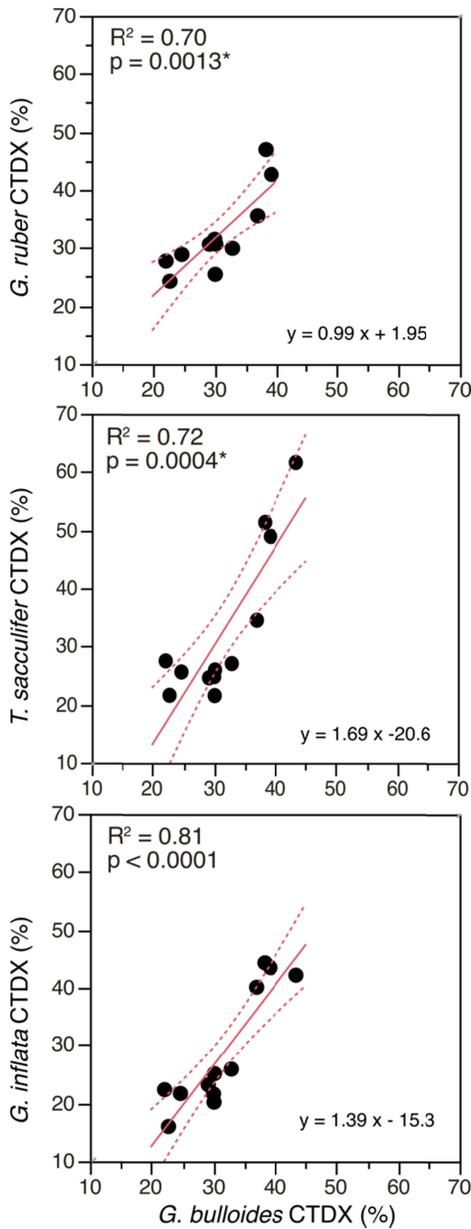
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 289 Figure 4. Changes in cross-sectional isosurface images and CT number histograms of four major
 290 species of planktonic foraminiferal shells (*G. bulloides*, *G. inflata*, *G. ruber* and *T. sacculifer*)
 291 with the progression of dissolution. Shells were obtained from three core top samples (GeoB-
 292 1728, 1729, and 3827). The condition of selected tests, which have CTDx similar to average of
 293 each sample set, are shown. $\Delta[\text{CO}_3^{2-}]$ at the nearest GLODAP stations at the core-top sample
 294 depths ranged from -4.3 to $23.8 \mu\text{mol kg}^{-1}$. The dashed line in the CT number histograms shows
 295 the threshold (50) between low and high values.

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 299 Figure 5. Inter species comparison of *G. bulloides* CTDX with other three species obtained from
 300 same core-top samples. The square of the correlation coefficient (R^2) and the type I error rate (p -
 301 value) are also shown.

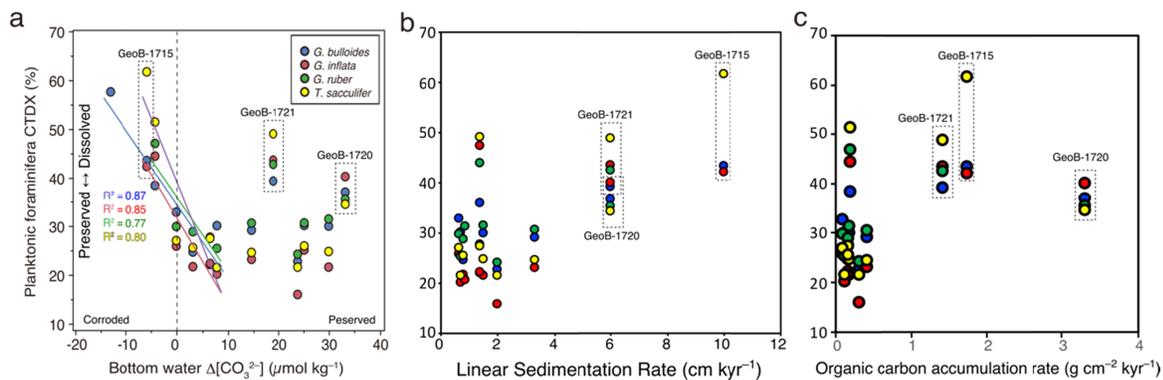
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303 **3.2. Comparison between planktonic foraminiferal CTDX, bottom water $\Delta[\text{CO}_3^{2-}]$, MAR,
 304 and organic carbon flux**

305 To assess the main controlling factors on foraminifera shell dissolution on the deep seafloor,
 306 we compared the planktonic foraminiferal shell CTDX values as a carbonate dissolution index,
 307 with three parameters: Bottom water $\Delta[\text{CO}_3^{2-}]$ representing ambient seawater corrosiveness,
 308 Linear Sedimentation Rate describing the exposure time of the shells on the seafloor, and
 309 Organic carbonate accumulation rate approximating supralysoclineal dissolution due to elevated
 310 porewater CO_2 from organic matter remineralization, for the 13 Atlantic samples where all
 311 parameters were available (Figure 6). First, we found a significant correlation between the

312 planktonic foraminiferal CTDX and the bottom water $\Delta[\text{CO}_3^{2-}]$ under the low $\Delta[\text{CO}_3^{2-}]$
 313 condition ($< 10 \mu\text{mol kg}^{-1}$): *G. bulloides*; $R^2 = 0.87$, *G. inflata*; $R^2 = 0.85$, *G. ruber*; $R^2 = 0.77$, *T.*
 314 *sacculifer*; $R^2 = 0.80$ (Figure 6a). In contrast, the comparison between CTDX and sedimentation
 315 rate site did not show any support for the hypothesis that higher sedimentation rate facilitates
 316 better preservation of carbonate by faster burial (Figure 6b). These results suggest that the
 317 carbonate dissolution intensity in the South Atlantic is generally governed by bottom water
 318 $\Delta[\text{CO}_3^{2-}]$ under low $\Delta[\text{CO}_3^{2-}]$ condition ($< 10 \mu\text{mol kg}^{-1}$), not by sedimentation rate. Under high
 319 $\Delta[\text{CO}_3^{2-}]$ conditions ($> 10 \mu\text{mol kg}^{-1}$), high CTDX values are observed at two sites (GeoB 1720
 320 and 1721), suggesting significant dissolution of carbonate, despite the supersaturated conditions
 321 for calcite in the ambient seawater. Here, a comparison with organic carbon accumulation rate (g
 322 $\text{cm}^{-2} \text{kyr}^{-1}$) reveals that the organic carbon inputs at both sites is more than three times higher
 323 than at the other sites (Figure 6c). These results indicate that organic carbon decomposition in the
 324 seafloor sediment may affect preservation of planktonic foraminifera shells even under
 325 supralysoclinal conditions ($> 10 \mu\text{mol kg}^{-1}$).

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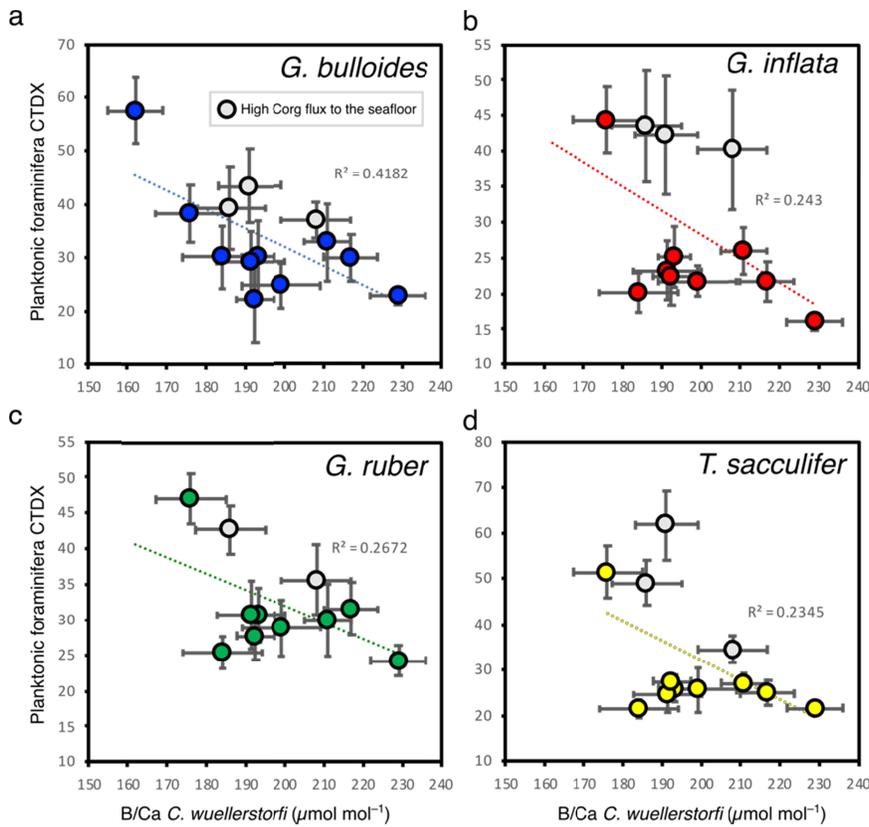
328
329 Figure 6. Investigation in controlling factor of foraminiferal shell dissolution intensity. (a)
 330 Bottom water $\Delta[\text{CO}_3^{2-}]$ ($\mu\text{mol kg}^{-1}$), (b) Linear Sedimentation Rate (cm kyr^{-1}), and (c) Organic
 331 carbon accumulation rate ($\text{g cm}^{-2} \text{kyr}^{-1}$) versus Planktonic foraminifera CTDX (%) at each
 332 sampling site. The data points at the sites locate in the Benguela upwelling system (GeoB-1715,
 333 1720 and 1721) with high organic carbon accumulation rate are surrounded by dotted line.

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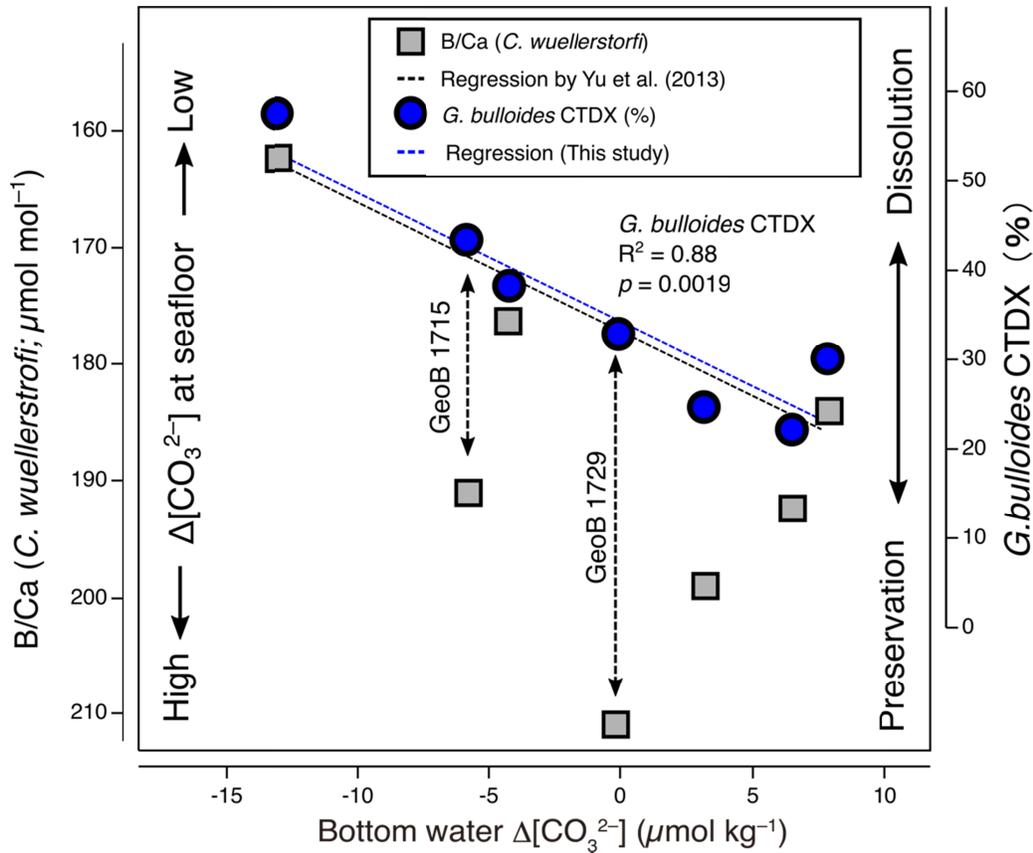
335 3.3. Comparison with the proxy of benthic foraminiferal B/Ca ratio

336 To investigate the applicability of the CTDX as a deep seawater $\Delta[\text{CO}_3^{2-}]$ proxy, we next
 337 compared the CTDX values in the 13 South Atlantic samples with the B/Ca ratio of benthic
 338 foraminifera (*C. wuellerstorfi*) (Figure 7). The B/Ca ratio of benthic foraminifera (*C.*
 339 *wuellerstorfi*), a conventional quantitative proxy of deep seawater $\Delta[\text{CO}_3^{2-}]$, derived from the
 340 same sediment samples showed a weak correlation with the CTDX of each species: *G. bulloides*;
 341 $R^2 = 0.42$, $p = 0.0169^*$, *G. inflata*; $R^2 = 0.24$, $p = 0.1034$, *G. ruber*; $R^2 = 0.27$, $p = 0.1034$, *T.*
 342 *sacculifer*; $R^2 = 0.28$, $p = 0.0960$ (Figure 7). The correlation has in all cases the expected sign
 343 (both proxies indicating more dissolution in the same samples), but the observation that the
 344 CTDX and B/Ca ratio correlation is not as pronounced as the correlation between CTDX and the
 345 bottom water $\Delta[\text{CO}_3^{2-}]$ is puzzling. Next, we focused on the *G. bulloides* CTDX, because this
 346 species was found at all sites, and showed the most significant correlation with the benthic
 347 foraminiferal B/Ca ratio among four species. The Figure 8 shows the relationship between two

348 proxies (*G. bulloides* CTDX and benthic foraminifera B/Ca ratio) and the bottom water $\Delta[\text{CO}_3^{2-}]$
 349 under the lower deep seawater $\Delta[\text{CO}_3^{2-}]$ condition ($< 10 \mu\text{mol kg}^{-1}$). Based on this comparison,
 350 we identified the sites where either proxy is discrepant, and investigated the factors that may be
 351 responsible for the discrepancy in these two proxies. In this plot, the regression lines between *G.*
 352 *bulloides* CTDX and bottom water $\Delta[\text{CO}_3^{2-}]$ obtained from the results of this study and that
 353 between the benthic foraminiferal B/Ca ratio and bottom water $\Delta[\text{CO}_3^{2-}]$ obtained from the
 354 results of calibration using more than 200 core-top samples (Yu et al., 2013) were presented. The
 355 result shows that the benthic foraminiferal B/Ca ratio is at least $20 \mu\text{mol mol}^{-1}$ higher than the
 356 regression line at two sites: Site 1715 (Depth: 4097 m, bottom water $\Delta[\text{CO}_3^{2-}]$: $-5.87 \mu\text{mol kg}^{-1}$);
 357 Site 1729 (Depth: 4401 m, bottom water $\Delta[\text{CO}_3^{2-}]$: $-0.10 \mu\text{mol kg}^{-1}$). At those sites, the
 358 estimated value of bottom water $\Delta[\text{CO}_3^{2-}]$ derived from the benthic foraminiferal B/Ca ratio
 359 imply higher than observed $\Delta[\text{CO}_3^{2-}]$.
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 364 Figure 7. Relationship between planktonic foraminiferal CTDX of (a) *G. bulloides*, (b) *G.*
 365 *inflata*, (c) *G. ruber*, and (d) *T. sacculifer* and B/Ca ratio of *C. wuellerstorfi* ($\mu\text{mol mol}^{-1}$) in each
 366 core-top sample. The sampling sites located in the Benguela upwelling system with high organic
 367 carbon flux are shown by gray circle. The regression line and the square of the correlation
 368 coefficient (R^2) are also shown.
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373 Figure 8. Comparison of the plots of *G. bulloides* CTDX and B/Ca ratio of *C. wuellerstorfi*
374 against bottom seawater $\Delta[\text{CO}_3^{2-}]$ under the condition of low bottom seawater $\Delta[\text{CO}_3^{2-}]$ (< 10
375 $\mu\text{mol kg}^{-1}$). The B/Ca ratio of *C. wuellerstorfi* at the site GeoB-1715, 1729 and 3803 are
376 relatively higher than *G. bulloides* CTDX.

377

3.4. Extended calibration between CTDX and deep water $\Delta[\text{CO}_3^{2-}]$

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380 In addition to the 13 core top samples collected in the South Atlantic, we added the 16 core
381 top samples collected in the Pacific Ocean to assess the behavior of the CTDX proxy under the
382 condition of low deep seawater $\Delta[\text{CO}_3^{2-}]$ ($< 10 \mu\text{mol kg}^{-1}$) and to obtain an extended
383 calibration equation between CTDX and deep seawater $\Delta[\text{CO}_3^{2-}]$ in the wider range of global
384 ocean. The plot shows the relationship between the CTDX of three species of major planktic
385 foraminiferal shell (*G. bulloides*, *G. ruber* and *T. sacculifer*) and deep water $\Delta[\text{CO}_3^{2-}]$ obtained
386 from nearby bottle water sampling at each site (Figure 9). The sites at GeoB 1720 and 1721,
387 with high organic material deposition, are excluded from the plot. The plot provides the
388 relationship between CTDX and a wide range of deep seawater $\Delta[\text{CO}_3^{2-}]$ (-27 to $30 \mu\text{mol kg}^{-1}$),
389 and suggests that CTDX of each species significantly correlate with deep seawater $\Delta[\text{CO}_3^{2-}]$
390 between the range of -20 to $10 \mu\text{mol kg}^{-1}$. The results of calibrations in each species are as
391 follows:

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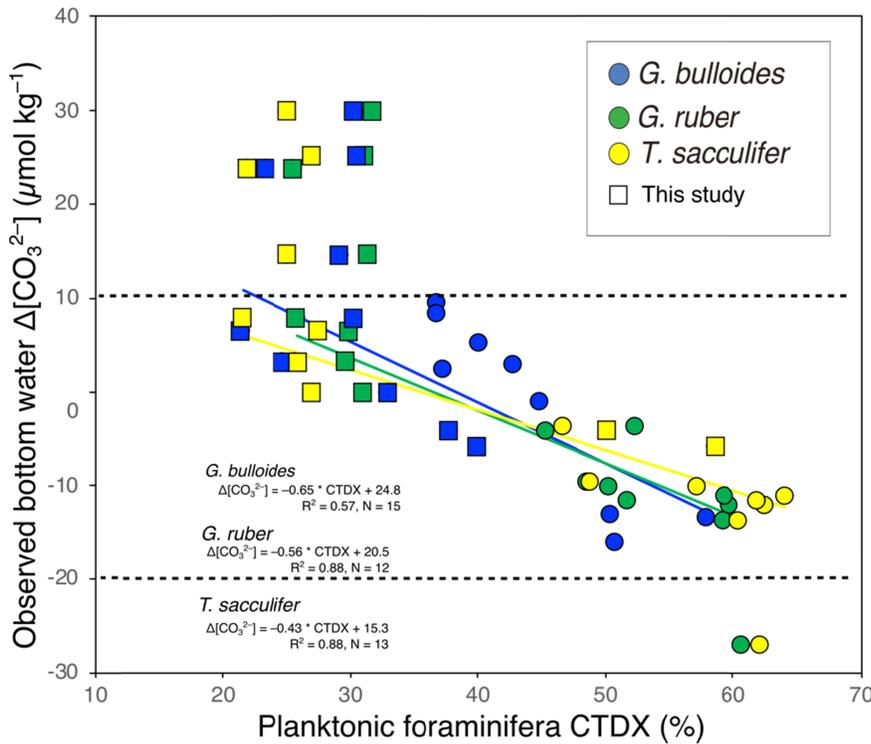
392 *G. bulloides*: Deep seawater $\Delta[\text{CO}_3^{2-}] = -0.65 * \text{CTDX} + 24.8$ ($R^2 = 0.57$, $N = 15$)

393 *G. ruber*: Deep seawater $\Delta[\text{CO}_3^{2-}] = -0.56 * \text{CTDX} + 20.5$ ($R^2 = 0.88$, $N = 12$)

394 *T. sacculifer*: Deep seawater $\Delta[\text{CO}_3^{2-}] = -0.43 * \text{CTDX} + 15.3$ ($R^2 = 0.88$, $N = 13$)

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These calibrations revealed that the CTDX of planktonic foraminifera are able to works as deep-water $[\text{CO}_3^{2-}]$ proxy in the specific range of the deep seawater $\Delta[\text{CO}_3^{2-}]$ (-20 to $10 \mu\text{mol kg}^{-1}$). However, the variation in the slope of regression line for each species indicated that the sensitivity of CTDX to dissolution varied among species. Based on the regression analysis, the uncertainties of the above regression equations are $\pm 5.7 \mu\text{mol kg}^{-1}$ (*G. bulloides*, $N = 15$), $\pm 2.8 \mu\text{mol kg}^{-1}$ (*G. ruber*, $N = 12$), and $\pm 2.6 \mu\text{mol kg}^{-1}$ (*T. sacculifer*, $N = 13$). On the other hand, the CTDX shows stable values of around 25 and 60 under the high ($> 10 \mu\text{mol kg}^{-1}$) and low ($< -20 \mu\text{mol kg}^{-1}$) carbonate saturation state condition, respectively.



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Figure 9. The CTDX of planktonic foraminifera (*G. bulloides*, *G. ruber* and *T. sacculifer*) in core top samples plotted against deep-water $\Delta[\text{CO}_3^{2-}]$. Core-top samples are obtained from this study and previous studies (Iwasaki et al., 2019 and 2022).

411 **4 Discussion**

412 **4.1. Controlling factors of the planktonic foraminiferal CTDX**

413 Dissolution of the planktonic foraminiferal shell is considered to occur mainly on deep
414 seafloor under the control of ambient seawater $\Delta[\text{CO}_3^{2-}]$. At the same time, it can be affected by
415 the variance in sedimentation rate and organic carbon respiration at the sediment surface [e.g.,
416 Berger et al., 1982, Milliman, 1993]. The direct comparisons between three numerical indexes
417 derived from deep seafloor in this study (i.e., planktic foraminiferal CTDX, bottom water
418 $\Delta[\text{CO}_3^{2-}]$, the sedimentation rate, and the organic carbon accumulation rate at each site) provided
419 the critical information to identify the controlling factor of carbonate dissolution at the deep
420 seafloor. First, the significant correlation between the proxy of CTDX and bottom water $\Delta[\text{CO}_3^{2-}]$

421] proves that dissolution of the planktic foraminiferal shell is undoubtedly controlled by bottom
 422 water $\Delta[\text{CO}_3^{2-}]$ in the most sites under the lower condition of bottom water $\Delta[\text{CO}_3^{2-}]$ ($< 10 \mu\text{mol}$
 423 kg^{-1}). Second, the sedimentation rate is not correlated with the planktic foraminiferal CTDX.
 424 This result suggests that the variance in sedimentation rate, which corresponds to approx. 5-to-
 425 10-fold difference in sedimentation rate does not significantly impact the carbonate dissolution
 426 variance at the deep seafloor. Third, the CTDX showed significantly higher values at two data
 427 sites (GeoB-1720 and 1721) than at the other data points under the condition of higher deep
 428 seawater $\Delta[\text{CO}_3^{2-}]$ ($> 10 \mu\text{mol kg}^{-1}$), which implies the exceeded carbonate dissolution under the
 429 supersaturated condition. These two sites are located at the Benguela Upwelling System in the
 430 eastern South Atlantic, one of the significant continental margins upwelling systems and belongs
 431 to the high productivity area of the ocean [Berger, 1989]. Previous research in the Benguela
 432 Upwelling System suggested that carbonate dissolution occurs at 400-1600 m above the
 433 lysocline by the above process (Vobers and Hinrich, 2002), induced by the release of CO_2 due to
 434 the oxidation of organic material within the sediment (Emerson and Bender, 1981; Archer, 1991;
 435 Jahnke et al., 1994). In addition, the other potential factor is that the porous shell, possibly
 436 response to rapid chamber formation, resulting in the formation of shells that are prone to
 437 dissolution. Culture studies have shown that the chamber formation rate increases under the food
 438 abundant condition as like as upwelling region (Bé et al., 2009), and such high chamber
 439 formation rate could provide the porous shell formation (Berger, 1970). In either case, we
 440 consider that the observed dissolution of planktonic foraminifera shells in the sediments derived
 441 from the Benguela Upwelling System in this study is not controlled by deep seawater $\Delta[\text{CO}_3^{2-}]$,
 442 but was probably brought by other factors such as decomposition of organic material occurring
 443 above the regional lysocline and ontogenetic growth condition at sea surface.

444 On the other hand, the relationship between benthic foraminiferal B/Ca ratio and deep-
 445 water $\Delta[\text{CO}_3^{2-}]$ did not show any unusual values at the sites from the Benguela Upwelling
 446 System, suggesting that the organic material decomposition does not have notable impact on
 447 benthic foraminiferal B/Ca ratio. We suppose that the reduction in seawater $\Delta[\text{CO}_3^{2-}]$, caused by
 448 organic material decomposition, has a more substantial influence on the dissolution of the
 449 planktonic foraminiferal test than the reduction in the B/Ca ratio of the benthic foraminiferal test.
 450 The mechanism to explain these results is to hypothesize that the decrease in seawater $\Delta[\text{CO}_3^{2-}]$
 451 due to organic material decomposition occurs mainly inside the sediment or aggregates of
 452 organic material like marine snow or fecal pellets. Such organic material oxidization affects
 453 planktonic foraminifera shells co-deposited with such aggregates, but, it does not significantly
 454 affect the seafloor surface or outside of the aggregates (i.e., the habitat area of benthic
 455 foraminifera). These findings suggests that the planktonic foraminiferal CTDX is not a suitable
 456 proxy at sites with high organic material deposition like upwelling system. The benthic
 457 foraminiferal B/Ca ratio is rather appropriate for use in such sample sites.

458 459 460 **4.2. Deep seawater $[\text{CO}_3^{2-}]$ proxies: The characteristics of CTDX and B/Ca proxy**

461 In this study, the characteristics of CTDX proxy were determined by measuring the
 462 dissolution intensity of four species of planktic foraminifera and comparing the several
 463 parameters derived from seafloor sediment samples in the South Atlantic. First, the results of
 464 inter-species comparisons (Figure 5) showed that the CTDX varies among the species,
 465 suggesting that species-specific application is required for accurate reconstruction. Nevertheless,
 466 our results suggested that the CTDX of all species are applicable as shell dissolution proxy. This

467 indicates that multiple species of planktonic foraminifera are applicable as deep-water $[\text{CO}_3^{2-}]$
468 proxy by establishing proxy calibration for each species. This is one of the essential advantages
469 of using CTDX over conventional proxies based on the B/Ca ratio of benthic foraminifera.
470 Therefore, it enables us to apply this new proxy to the foraminifera poor sediment cores, where it
471 is challenging to collect single species continuously. It also allows us to compare the
472 reconstructed data between sites from different locations with different planktic locations
473 foraminiferal species compositions.

474 Second, the relationship between the CTDX and the deep seawater $\Delta[\text{CO}_3^{2-}]$ suggested
475 that the CTDX significantly correlated with deep seawater $\Delta[\text{CO}_3^{2-}]$ only under the low deep
476 seawater $\Delta[\text{CO}_3^{2-}]$ condition ($< 10 \mu\text{mol kg}^{-1}$) (Figure 6, and Figure 9). However, under the
477 high deep seawater $\Delta[\text{CO}_3^{2-}]$ condition ($> 10 \mu\text{mol kg}^{-1}$), the values of planktic
478 foraminiferal CTDX were almost constant. Similar results were also represented by the
479 comparison between shell dissolution proxy and deep seawater $\Delta[\text{CO}_3^{2-}]$, suggesting the
480 saturation of shell dissolution proxy at the sites of high deep seawater $\Delta[\text{CO}_3^{2-}]$ (> 10
481 $\mu\text{mol/kg}$) [Johnstone et al., 2010]. Furthermore, under the extremely low deep seawater
482 $\Delta[\text{CO}_3^{2-}]$ condition ($< -20 \mu\text{mol kg}^{-1}$), the CTDX seems to saturate around the values of 60
483 (Figure 9). This implies that planktonic foraminiferal shells which appear intact (i.e., which
484 are not broken or fragmented and retain the outermost chamber) show maximum CTDX
485 values of around 60. It is likely that with further progressing dissolution the shells lose
486 structural integrity, break into fragments and would fall outside of the sample selection for
487 CT scanning. From the above, we conclude that the CTDX proxy works effectively under
488 deep seawater $\Delta[\text{CO}_3^{2-}]$ lower than $10 \mu\text{mol kg}^{-1}$ and higher than $-20 \mu\text{mol kg}^{-1}$.
489 Alternatively, we suppose that the analysis of fragmented individual is useful for applying
490 CTDX under the significantly low deep seawater $\Delta[\text{CO}_3^{2-}]$ condition ($< -20 \mu\text{mol kg}^{-1}$).

491 We also evaluated the accuracy of the deep seawater $[\text{CO}_3^{2-}]$ reconstruction by each
492 proxy. Under the low deep seawater $\Delta[\text{CO}_3^{2-}]$ ($< 10 \mu\text{mol/kg}$) condition, the proxy of CTDX
493 showed a correlation with the deep seawater $\Delta[\text{CO}_3^{2-}]$ and the conventional proxy of benthic
494 foraminiferal B/Ca ratio (Figure 7). These results support our interpretation that the proxy of
495 CTDX can serve as an indicator of deep seawater $\Delta[\text{CO}_3^{2-}]$. Nevertheless, comparing these two
496 proxies, we found two peculiar sites (Sites GeoB-1715 and 1729) where the CTDX and B/Ca
497 ratio values deviate (Figure 8). The sediment samples at these two sites are rich in carbonate ($>$
498 80%), and the proxy of CTDX shows the occurrence of carbonate dissolution and relatively high
499 values of benthic foraminiferal B/Ca ratio than general estimation. We speculate that the process
500 of carbonate compensation in surface sediment may have raised $[\text{CO}_3^{2-}]$ in the porewater and
501 contributed to the higher values of the benthic foraminiferal B/Ca ratio in a way similar to the
502 speculated protective effect of aragonite dissolution on carbonate preservation described by
503 Sulpice et al. (2022). To evaluate this speculation, we investigated the difference in total
504 alkalinity of porewater in the sediment surface (upper 5 cm) and the overlying bottom water
505 (Hensen et al., 2003a-d). The total alkalinity is a valuable indicator that relates to the seawater
506 $[\text{CO}_3^{2-}]$ and is controlled by the carbonate dissolution in the sediment, which enables us to
507 presume the geochemical process in the sediment surface. The profiles of chemical parameters at
508 Site GeoB-1715 and 1729 showed that the total alkalinity in the pore water (average of the upper
509 5 cm) is more than $200 \mu\text{mol kg}^{-1}$ higher than in overlying bottom water as follows: GeoB-1715
510 (bottom water: $2354 \mu\text{mol kg}^{-1}$, porewater average: $2583 \mu\text{mol kg}^{-1}$) and GeoB-1729 (bottom
511 water: $2359 \mu\text{mol kg}^{-1}$, porewater average: $2653 \mu\text{mol kg}^{-1}$). In contrast, at Site GeoB-3804 and
512 3827 with appropriate benthic foraminiferal B/Ca ratio for the general estimation, the difference

513 in the total alkalinity between pore water and overlying bottom water is relatively small as
514 follows: GeoB-3804 (bottom water: 2348 $\mu\text{mol kg}^{-1}$, porewater average: 2458 $\mu\text{mol kg}^{-1}$) and
515 GeoB-3827 (bottom water: 2342 $\mu\text{mol kg}^{-1}$, porewater average: 2275 $\mu\text{mol kg}^{-1}$). These results
516 imply that carbonate dissolution at the sediment surface discharges carbonate ions and raises the
517 porewater's total alkalinity, altering the calcification condition for benthic foraminifera. From the
518 above, we consider that the strong carbonate dissolution in the carbonate-rich sediment is one of
519 the principles controlling factors of the benthic foraminiferal B/Ca ratio.

520

521 5 Conclusions

522 The assessments in planktic foraminiferal shell dissolution intensity, represented by
523 CTDX, were directly compared with the data of B/Ca in benthic foraminifera in the same sample
524 and the other proxies of conditions at each site (deep-water $\Delta[\text{CO}_3^{2-}]$, sedimentation rate, and
525 Organic carbon accumulation rate). In addition, inter-species variation in CTDX of four species
526 of planktic foraminifera were assessed. Our results proved that CTDX of each planktic
527 foraminifera is generally controlled by deep-water $\Delta[\text{CO}_3^{2-}]$ variation like a conventional proxy
528 of benthic foraminiferal B/Ca ratio. We suggested that four species (*G. bulloides*, *G. inflata*, *G.*
529 *ruber*, and *T. sacculifer*) of planktonic foraminiferal CTDX are applicable as a quantitative proxy
530 for carbonate dissolution intensity, while we recommend species-specific use of CTDX for
531 accurate reconstruction because there is slight difference in sensitivity to dissolution. On the
532 other hand, the effect of variation in sedimentation rate is not a significant factor in carbonate
533 dissolution in the sediment surface if the difference in sedimentation rate is 5-to-10-fold
534 difference or less. Based on the above results, we concluded that the proxy of planktic
535 foraminiferal CTDX is applicable as a quantitative deep-water $[\text{CO}_3^{2-}]$ proxy as well as a
536 conventional proxy of benthic foraminiferal B/Ca ratio. Nevertheless, there are several caveats in
537 the application of each proxy. First, the proxy of CTDX is useful under the specific condition of
538 deep-water $\Delta[\text{CO}_3^{2-}]$ between -20 to $10 \mu\text{mol kg}^{-1}$, suggesting that the CTDX has a detection
539 limit range of deep-water $[\text{CO}_3^{2-}]$. Therefore, we should exclude the seafloor sediment sample
540 obtained from high deep-water $\Delta[\text{CO}_3^{2-}]$ condition ($> 10 \mu\text{mol kg}^{-1}$) in the process of
541 establishing the calibration equation, and we have to note that this proxy is not reliable for
542 reconstructing the higher deep-water $\Delta[\text{CO}_3^{2-}]$ in the application of CTDX into sediment core
543 samples. Second, we also observed the excessive dissolution of planktic foraminifera in the
544 supersaturated condition for calcite, which seems to be caused by the intensive input of organic
545 material into seafloor sediment. Therefore, we consider that we should refrain from applying the
546 CTDX proxy to the sediment sample located in the upwelling system where organic material
547 input from the sea surface is exceptionally high. Third, we revealed that carbonate compensation,
548 which occurs in carbonate-rich and carbonate dissolution affected sediment samples, may alter
549 the benthic foraminiferal B/Ca ratio higher during their calcification, which may contribute to
550 uncertainty of deep-water $[\text{CO}_3^{2-}]$ reconstruction by benthic foraminiferal B/Ca ratio.

551 Based on the suggested characteristics of CTDX, we provided the calibration formula
552 between three species planktic foraminiferal (*G. bulloides*, *G. ruber* and *T. sacculifer*) CTDX
553 and deep seawater $\Delta[\text{CO}_3^{2-}]$ using a number ($N = 22$) of seafloor sediment samples. Although
554 the number of sediment samples and the distribution range of samples are inferior to the
555 conventional proxy of benthic foraminiferal B/Ca ratio, our calibration showed that the proxy of
556 CTDX works as well as the B/Ca proxy for deep seawater $[\text{CO}_3^{2-}]$ reconstruction under the
557 suitable conditions. This proxy may help to fill the blank of paleo-deep-water $[\text{CO}_3^{2-}]$ data, in

558 particular, in the mid-high latitude of the North Pacific, where benthic foraminiferal B/Ca ratio
559 data are insufficient.

560

561

562 **Acknowledgments**

563 This study used samples that were collected during cruises M12/1, M20/2 and M34/3 by R/V
564 METEOR. The data reported in this paper are available in Supplemental Information. This study
565 was technically and financially supported by the Japan Agency for Marine-Earth Science and
566 Technology and the Cluster of Excellence "The Ocean Floor—Earth's Uncharted Interface"
567 funded by the German Research Foundation (DFG).

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570 **Open Research**

571 The data used in this study will be available in PANGAEA (<https://www.pangaea.de/>).

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760

761 **Figure captions**

762 **Figure 1.** The locations of multiple corer samples in the South Atlantic used in this study. The
763 section along the yellow line is showing the spatial distribution of the degree of carbonate
764 saturation ($\Delta[\text{CO}_3^{2-}]$) in the study area.
765

766 **Figure 2.** B/Ca ratio in shells of the benthic foraminifera *Cibicidoides wuellerstorfi* in core top
767 samples plotted against deep-water $\Delta[\text{CO}_3^{2-}]$. The data of B/Ca ratio of benthic foraminifera in
768 the core-top sample used in this study are shown by blue squares. Data are derived from [Rae et](#)
769 [al. \(2011\)](#), [Raitzch et al. \(2011\)](#), and [Yu and Elderfield \(2007\)](#).
770

771 **Figure 3.** Protocol to calculate the foraminiferal shell dissolution index (CTDX) start from raw
772 sequential X-ray image data (a). (b) Contaminants in the shell sample are removed by manual
773 deleting. (c) Apply threshold to distinguish between calcite material and surrounding
774 environment. Extract CT number histogram from cleaned and threshold 3D data. (d) Scaling X
775 and Y-axes of CT number histogram. (e) Calculation of CTDX (%).

776

777 **Figure 4.** Changes in cross-sectional isosurface images and CT number histograms of four major
 778 species of planktonic foraminiferal shells (*G. bulloides*, *G. inflata*, *G. ruber* and *T. sacculifer*)
 779 with the progression of dissolution. Shells were obtained from three core top samples (GeoB-
 780 1728, 1729, and 3827). The condition of selected tests, which have CTDX similar to average of
 781 each sample set, are shown. $\Delta[\text{CO}_3^{2-}]$ at the nearest GLODAP stations at the core-top sample
 782 depths ranged from -4.3 to $23.8 \mu\text{mol kg}^{-1}$. The dashed line in the CT number histograms shows
 783 the threshold (50) between low and high values.

784

785 **Figure 5.** Inter species comparison of *G. bulloides* CTDX with other three species obtained from
 786 same core-top samples. The square of the correlation coefficient (R^2) and the type I error rate (p -
 787 value) are also shown.

788

789 **Figure 6.** Investigation in controlling factor of foraminiferal shell dissolution intensity. (a)
 790 Bottom water $\Delta[\text{CO}_3^{2-}]$ ($\mu\text{mol kg}^{-1}$), (b) Linear Sedimentation Rate (cm kyr^{-1}), and (c) Organic
 791 carbon accumulation rate ($\text{g cm}^{-2} \text{ kyr}^{-1}$) versus Planktonic foraminifera CTDX (%) at each
 792 sampling site. The data points at the sites locate in the Benguela upwelling system (GeoB-1715,
 793 1720 and 1721) with high organic carbon accumulation rate are surrounded by dotted line.

794

795 **Figure 7.** Relationship between planktonic foraminiferal CTDX of (a) *G. bulloides*, (b) *G.*
 796 *inflata*, (c) *G. ruber*, and (d) *T. sacculifer* and B/Ca ratio of *C. wuellerstorfi* ($\mu\text{mol mol}^{-1}$) in each
 797 core-top sample. The sampling sites located in the Benguela upwelling system with high organic
 798 carbon flux are shown by gray circle. The regression line and the square of the correlation
 799 coefficient (R^2) are also shown.

800

801 **Figure 8.** Comparison of the plots of *G. bulloides* CTDX and B/Ca ratio of *C. wuellerstorfi*
 802 against bottom seawater $\Delta[\text{CO}_3^{2-}]$ under the condition of low bottom seawater $\Delta[\text{CO}_3^{2-}]$ (< 10
 803 $\mu\text{mol kg}^{-1}$). The B/Ca ratio of *C. wuellerstorfi* at the site GeoB-1715, 1729 and 3803 are
 804 relatively higher than *G. bulloides* CTDX.

805

806 **Figure 9.** The CTDX of planktonic foraminifera (*G. bulloides*, *G. ruber* and *T. sacculifer*) in
 807 core top samples plotted against deep-water $\Delta[\text{CO}_3^{2-}]$. Core-top samples are obtained from this
 808 study and previous studies (Iwasaki et al., 2019 and 2022).

809

810 **Table 1.** Multiple core sample locations, depth (m), Deep water degree of $\Delta[\text{CO}_3^{2-}]$ ($\mu\text{mol kg}^{-1}$),
 811 Linear Sedimentation Rate (cm kyr^{-1}), TOC (Total Organic Carbon) accumulation rate ($\text{g m}^{-2} \text{ y}^{-1}$)
 812 and *Cibicidoides wuellerstorfi* B/Ca ratio ($\mu\text{mol mol}^{-1}$) at each sampling site.

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