

High-precision measurement of $^{36}\text{SF}_5^+$ signal using the MAT 253 Ultra isotope-ratio mass spectrometer

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Abstract

Rationale: The $\Delta^{36}\text{S}$ standard deviation (SD) measured in a conventional isotope-ratio mass spectrometer (IRMS) such as MAT 253 is at *c.a.* 0.1‰ difficult to resolve the origin of non-mass-dependent sulfur (NMD-S) isotope fractionation in the tropospheric sulfate aerosol and in Martian meteorites or small deviations from the canonical mass-dependent fractionation laws. Interfering ions (originated from fluorination and/or mass spectrometer ion source itself) with m/z at 131 of $^{36}\text{SF}_5^+$ by the community as the cause of the poor precision, but the exact ion species has not been identified or confirmed. **Methods:** Here we examined the potential interfering ions by using the Thermo Scientific MAT 253 Ultra, a high-resolution (mass resolving powers up to 40,000) stable isotope-ratio mass spectrometer, to measure the SF_6 working gas and SF_6 gases converted from IAEA-S1 Ag_2S reference materials via a fluorination system. **Results:** We found that there is a resolvable peak to the right of the $^{36}\text{SF}_5^+$ peak for both the SF_6 working gas and SF_6 generated by the fluorination system. The peak is identified as the $^{12}\text{C}_3\text{F}_5^+$ ion, generated inside the instrument during the ionization process. By minimizing the presence of carbon-bearing compounds (e.g., organic matter in sample or helium gas, glue or o-ring in sample tubes, or carbon-bearing gases inside the mass spectrometer), we were able to achieve a $\Delta^{36}\text{S}$ SD of 0.046‰ SF_6 zero-enrichment and 0.069‰ measurement start from sliver sulfide IAEA-S1. **Conclusions:** Minimizing the presence of carbon-bearing compounds and avoiding the interfering signals from $^{36}\text{SF}_5^+$ via MAT 253 Ultra high-resolution isotope-ratio mass spectrometer, we can improve $\Delta^{36}\text{S}$ measurement precision by 2 to 5 folds, which helps to open new territories for research using quadruple sulfur isotope composition.

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